

Study of a large scale powdered activated carbon pilot: Removals of a wide range of emerging and priority micropollutants from wastewater treatment plant effluents

R. Mailler, J. Gasperi, Y. Coquet, S. Deshayes, S. Zedek, C. Cren-Olivé, N. Cartiser, V. Eudes, Adèle Bressy, E. Caupos, et al.

► To cite this version:

R. Mailler, J. Gasperi, Y. Coquet, S. Deshayes, S. Zedek, et al.. Study of a large scale powdered activated carbon pilot: Removals of a wide range of emerging and priority micropollutants from wastewater treatment plant effluents. Water Research, 2015, 72, pp.315-330. 10.1016/j.watres.2014.10.047 . hal-01103115

HAL Id: hal-01103115 https://enpc.hal.science/hal-01103115

Submitted on 19 Jun 2018

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

1	Study of a large scale powdered activated carbon pilot: removals of a wide range of
2	emerging and priority micropollutants from wastewater treatment plant effluents
3	
4	R. Mailler ^{1*} , J. Gasperi ^{1*} , Y. Coquet ² , S. Deshayes ^{1,3} , S. Zedek ¹ , C. Cren-Olivé ⁴ , N. Cartiser ⁴ , V. Eudes ³ , A.
5	Bressy ¹ , E. Caupos ¹ , R. Moilleron ¹ G. Chebbo ⁵ and V. Rocher ⁶
6	
7	¹ LEESU (UMR MA 102, Université Paris-Est, AgroParisTech), 61 avenue du Général de Gaulle, 94010
8	Créteil Cedex, France. (E-mail: maillerr@leesu.enpc.fr; gasperi@u-pec.fr)
9	² SAUR, Direction de la Recherche et du Développement, 1 rue Antoine Lavoisier, 78064 Guyancourt,
10	France.
11	³ LCPP (Laboratoire Central de la Préfecture de Police), 39 bis rue de Dantzig, 75015 Paris, France.
12	⁴ ISA (UMR 5280, Institut des sciences analytiques, CNRS), 5 rue de la Doua, 69100 Villeurbanne, France.
13	⁵ LEESU (UMR MA 102, Université Paris-Est, AgroParisTech), 6-8 avenue Blaise Pascal, Champs-sur-
14	Marne, 77455 Marne-la-Vallée Cedex 2, France.
15	⁶ SIAAP, Direction du Développement et de la Prospective, 82 avenue Kléber, 92700 Colombes, France.
16	* Corresponding authors
17	
18	HIGHLIGHTS
19	• Pharmaceuticals and hormones (PhPHs) are well removed (> 60%) by powdered activated carbon (PAC)
20	• 50-95% removals are also achieved for parabens, bisphenol A and pesticides by PAC
21	• Treatment degradation leads to substantially lower PhPHs removals by PAC
22	• The fresh PAC dose is strongly correlated to the process efficacy
23	• Molecular charge, hydrophobicity and molecular weight have a great influence on the micropollutant fate
24	
25	

26 ABSTRACT

The efficacy of a fluidized powdered activated carbon (PAC) pilot (CarboPlus®) was studied in both nominal (total nitrification + post denitrification) and degraded (partial nitrification + no denitrification) configuration of the Seine Centre WWTP (Colombes, France). In addition to conventional wastewater parameters 54 pharmaceuticals and hormones (PhPHs) and 59 other emerging pollutants were monitored in influents and effluents of the pilot. Thus, the impacts of the WWTP configuration, the process operation and the physico-chemical properties of the studied compounds were assessed in this article.

33

Among the 26 PhPHs quantified in nominal WWTP configuration influents, 8 have high dissolved concentrations (>100 ng/L), 11 have an intermediary concentration (10-100 ng/L) and 7 are quantified below 10 ng/L. Sulfamethoxazole is predominant (about 30% of the sum of the PhPHs). Overall, 6 PhPHs are poorly to moderately removed (<60%), such as ibuprofen, paracetamol or estrone, while 9 are very well removed (>80%), i.e. beta blockers, carbamazepine or trimethoprim, and 11 are well eliminated (60-80%), i.e. diclofenac, naproxen or sulfamethoxazole.

In degraded WWTP configuration, higher levels of organic matter and higher concentrations of most pollutants are observed. Consequently, most PhPHs are substantially less removed in percentages but the removed flux is higher. Thus, the PAC dose required to achieve a given removal percentage is higher in degraded WWTP configuration. For the other micropollutants (34 quantified), artificial sweeteners and phthalates are found at particularly high concentrations in degraded WWTP configuration influents, up to µg/L range. Only pesticides, bisphenol A and parabens are largely eliminated (50-95%), while perfluorinated acids, PAHs, triclosan and sweeteners are not or weakly removed (<50%). The remaining compounds exhibit a very variable fate from campaign to campaign.

46

The fresh PAC dose was identified as the most influencing operation parameter and is strongly correlated to performances. Charge and hydrophobicity of compounds have been recognized as crucial for the micropollutant adsorption on PAC, as well as the molecular weight. Finally, a PAC dose of 10 mg/L allows an average removal of 72-80% of the sum of the PhPHs in nominal WWTP configuration. The comparaison of the results with those from the scarce other studies tends to indicate that an extrapolation of them to different PAC processes and to other WWTPs could be possible and relevant, taking into account the differences of water quality from WWTP to WWTP.

53

54 KEYWORDS

55 Pharmaceuticals; emerging pollutants; adsorption; powdered activated carbon; wastewater treatment

56

57 INTRODUCTION

58

In addition to priority pollutants (EC 2013), many other emerging micropollutants such as pharmaceuticals and hormones (PhPHs), personal care products (PCPs), pesticides, phthalates, artificial sweeteners, etc. are found in the aquatic environment (Jones et al. 2001, Lange et al. 2012, Luo et al. 2014). As bioactive and toxic substances, their environmental effects have been proven (Bolong et al. 2009, Daughton and Ternes 1999). For some of these compounds such as PhPHs and PCPs, wastewater treatment plant (WWTP) discharges have been identified as an important source in the aquatic environment (Halling-Sørensen et al. 1998, Heberer 2002).

65

66 Several studies have shown that conventional WWTPs (with primary and biological treatments) substantially remove 67 hydrophobic, volatile and biodegradable micropollutants even if they are not designed and/or optimized for that purpose 68 (Clara et al. 2007, Mailler et al. 2013, Ruel et al. 2012). However, hydrophilic or refractory organic compounds are still present in the treated wastewater at ng/L to µg/L level (Loos et al. 2013). Besides, emerging micropollutants, 69 70 particularly pharmaceuticals such as diclofenac, may be included in future modifications of existing European (EC 71 2013) or national regulations. Among the solutions to reduce the emerging contaminant discharges into the 72 environment, the implementation of a tertiary treatment in conventional WWTPs is more and more considered by 73 WWTP managers and decision makers. These technologies, such as membrane filtration, advanced oxidation and 74 adsorption processes, are principally adapted from drinking water industry. They have been employed for a long time, 75 but the generalization of their applications to wastewater treatment has been seriously considered recently. However, 76 very few studies have shown the potential of such technologies to remove organic pollutants from treated wastewater 77 (Boehler et al. 2012, Margot et al. 2013) and their economic feasibility (Abegglen and Siegrist 2012).

78

In this context, a large scale powdered activated carbon (PAC) pilot has been set up at the Seine Centre WWTP, treating
wastewater from Paris conurbation and supervised by the Parisian public sanitation service (SIAAP). An extended
sampling strategy, both in number of campaigns (n=14) and compounds (N=113), has been built.

82

Thus, this paper aims at displaying and comparing the performances of such a process for micropollutant removal in both nominal and degraded WWTP configurations, and to assess the influence of both operation parameters and physico-chemical properties of the pollutants. First, the influent dissolved concentrations for both micropollutants and

86 conventional wastewater parameters are discussed and compared to the literature. Then, the removals of the molecules are examined, as well as the effluent dissolved concentrations and the water quality. The difference between a normal 87 and a degraded configuration of the WWTP is particularly analyzed. Finally, the influence of operation parameters 88 89 (fresh PAC dose, total mass of carbon in the reactor, hydraulic velocity) and physico-chemical properties of the 90 compounds (charge, hydrophobicity, size, functional groups, etc.) are also examined. Such results are of high interest 91 and very relevant considering the poverty of the literature regarding the removal of micropollutants from wastewater by PAC. 92 93 94 MATERIALS AND METHODS 95 96 97 1. Site and process description 98 99 In nominal configuration, the Seine Centre WWTP treats 240 000 m³ per day of domestic wastewater from Paris city, 100 corresponding to 900 000 population equivalents. In these conditions (total nitrification + post denitrification), the 101 treatment process is composed by three major steps: pre, primary and biological treatments. After the pretreatment (screening, grit and oil removal units), a physicochemical lamellar settling unit (Densadeg[®]) removes a great part of the 102 particulate and colloidal pollution thanks to coagulant (ferric chloride) and flocculant (anionic polymer) addition. 103 104 Finally, a three stage biofiltration system achieves the biological treatment. In nominal configuration, the first stage 105 (Biofor® filters - biolite medium), allows the treatment of carbonaceous pollution in aerobic conditions, and the second (Biostyr[®] filters - biostyrene medium) and third (Biofor[®] filters) stages remove the nitrogenous pollution respectively in 106 aerobic (total nitrification) and anoxic (denitrification) conditions (Rocher et al. 2012). This WWTP is able to switch to 107 a degraded configuration to treat 405 000 m³ of water per day but a partial nitrification and no denitrification are 108 performed. A detailed layout of the WWTP in both configurations is given in supporting material - Figure S1. 109

110

The tertiary treatment process studied (CarboPlus[®] - Figure 1) is fed by Seine Centre treated effluents and treats between 700 and 1 200 m³/day. This pilot consists in a 5 m high reactor with a surface area of 4 m², where water flows upstream through a fluidized bed of PAC. Depending on the mass of PAC inside the reactor and the hydraulic velocity, the bed depth varies between 1 and 3 m. A dose of fresh PAC is continuously injected. At steady state, a comparable amount of PAC is regularly removed from the reactor to obtain a solid retention time (SRT) of 5 to 7 days, and an

116	overall concentration of 5 to 10 g/L of PAC is maintained inside the bed. In addition, coagulant (ferric chloride) and
117	flocculant (anionic polymer) are injected continuously with the fresh PAC to stabilize the bed expansion and to prevent
118	the leakage of PAC with treated water. The hydraulic velocity can technically vary from 6 to 12 m/h and the hydraulic
119	retention time considering the reactor size is comprised between 25 and 50 min. The observed contact time between the
120	PAC bed and the water varies between 10 and 20 min considering the bed depth and the hydraulic velocity.
121	FIGURE 1
122	
123	2. Study and sampling strategy
124	
125	The study has been initially built to assess the process efficacy in different operation configuration (Table 1) during 14
126	campaigns (from July to December 2013). However, the Seine Centre WWTP operated in degraded configuration
127	during the 5 lasts campaigns (unplanned), because it received additional water from another WWTP that was in
128	maintenance. Taking this into account, the campaigns performed have been organized in two phases in the results and
129	discussion sections of the article. The first 9 campaigns, corresponding to phase 1, were performed during nominal
130	WWTP configuration. The last 5 campaigns were carried out under degraded WWTP configuration, corresponding to
131	phase 2.
132	In addition, different process configurations were tested during the first phase (Table 1) by varying the fresh PAC dose
133	in the ranges operationally feasible within the process. Thus, three moderate PAC doses (5-10-20 mg/L) and one high
134	PAC dose (70 mg/L), as well as four hydraulic velocities (6-7-8-10 m/h) were tested, allowing determining their
135	influences on performances. The 70 mg/L of PAC campaign was performed to determine whether some compounds
136	could still be recalcitrant to adsorption at high dose. Information about operation parameters during this phase are given
137	in Table 1. In contrary, a stable and intermediate configuration (10 mg/L of PAC and 7 m/h), considered as the
138	recommended configuration of the process, was maintained during the second phase.
139	Thus, this experimental design allows assessing the influences of both the process operation parameters and the
140	configuration of the WWTP.
141	TABLE 1
142	
143	The same sampling procedure was applied for the 14 campaigns. Analyzes were performed on the dissolved phase. 24-h
144	composite samples of 20 L were collected using automatic refrigerated samplers (4°C) equipped with glass bottles,
145	properly cleaned, and Teflon [®] pipes to avoid any contamination or loss. When sampling was completed, glass bottles

146 were collected, very carefully homogenized and sub-samples were distributed for analyses. Each sample was filtered

147 using 0.7 µm GF/F filters (Whatman[®]). All samples were analyzed within 48-h after sampling, due to transport time to 148 the laboratories involved, except for sweeteners. As recommended by the laboratory performing the artificial 149 sweeteners analyses (internal tests), these samples were acidified (1% volumetric HCl) and stored in fridge (4°C) until 150 analyses.

151

152 **3.** Activated carbon characterization

153

Activated carbon characteristics are very important in the adsorption mechanisms and can strongly influence the 154 micropollutant fate (Cecen and Aktas 2012). Apart from BET (Brunauer, Emmett and Teller) specific surface and 155 particle size, the mesoporous structure is the best suited for micropollutants adsorption (Cecen and Aktas 2012, Ebie et 156 157 al. 2001), as it reduces the impact of organic matter competition with micropollutants. Thus, a mesoporous PAC 158 (DaCarb PB-170) was tested in the pilot, after preliminary lab tests comparing different PACs. The characteristics of this PAC are given in supporting material - Table 1. Globally, the selected PAC is featured by a high BET surface of 159 160 957 ± 28 g/m², which is close to BET specific surfaces found in the literature, often comprised between 700 and 1 500 m²/g (Cecen and Aktas 2012, de Ridder et al. 2013, Margot et al. 2013). In addition, its granulometry is large but its 161 162 median particle diameter is rather low ($<20 \mu m$), which theoretically enhance the adsorption kinetics. PAC observations 163 by scanning electron microscopy are provided in supporting material - Figure S2.

164

165 4. Pollutants and analytical procedures

166

In all samples, conventional wastewater parameters were analyzed by SIAAP French accredited laboratory (COFRAC, supporting material - Table S2) to characterize the general quality of water. These parameters include: dissolved organic carbon (DOC), chemical oxygen demand (COD), biological oxygen demand (BOD₅), UV absorption at 254 nm (UV-254), total Kjeldahl nitrogen (TKN), NH_4^+ , NO_3^- , NO_2^- , PO_4^{3-} , total phosphorous (TP) and total suspended solids (TSS). All analyses were performed according to standards (supporting material - Table S2).

172

173 113 micropollutants were monitored in the dissolved phase (Table 2). First, pharmaceuticals (N=44, N for the number 174 of compounds) and hormones (N=10) were analyzed for all campaigns, as they are the compounds targeted by the 175 process. In addition to these molecules, complementary analyses were performed for other micropollutants (N=59) 176 during phase 2. Thus, chlorinated solvents (N=3), perfluorinated acids (N=2), pesticides (N=23), bisphenol A,

177	alkylphenols (N=2) and phthalates (N=4) were analyzed in all campaigns of phase 2, while PAHs (N=13 - the 16 US-
178	EPA PAHs excluding naphthalene, acenaphthene and acenaphthylene) and PCPs (N=7) were monitored in 3 campaigns
179	and artificial sweeteners (N=4) in 4 campaigns. The full list of molecules is given in supporting material - Table S3, as
180	well as their respective limits of quantification (LQ).
181	
182	Micropollutant analyses were performed by 4 laboratories: the Institute of Analytical Sciences (ISA - Villeurbanne,

France) from CNRS, in charge of the PhPHs, chlorinated solvents, perfluorinated acids, pesticides and bisphenol A, the Water Environment and Urban Systems laboratory (LEESU - Créteil, France), in charge of PAHs and PCPs, the Central laboratory of the Police Prefecture (LCPP - Paris, France), in charge of phthalates and alkylphenols, and the Water Technology Center (TZW - Karlsruhe, Germany), in charge of artificial sweeteners.

187

TABLE 2

Information about the analytical procedures (Table 2) are available in the literature for PhPHs (Vulliet et al. 2011), pesticides and chlorinated solvents (Barrek et al. 2009), PAHs (Bressy et al. 2012), alkylphenols and phthalates (Bergé et al. 2014), sweeteners (Scheurer et al. 2009), and for PCPs (Gasperi et al. 2014).

191

192 5. Data processing

193

The statistical calculations were performed separately for the nominal (n=9) and the degraded (n=5) WWTP configurations, for comparison purposes. When the number of values available was higher than six, box plots were plotted with minimum, 1st quartile (Q1), 3rd quartile (Q3) and maximum values of the series. In contrary (≤ 6 values), individual values were plotted. Finally, the results given in text, concentrations as well as removals, are average results \pm standard deviations (>6 values) or minimum – maximum values (≤ 6 values).

199

As no direct assessment of the uncertainties was performed, and to have an idea of the robustness of the data, limit values of 5 times the LQ were defined for each compound. The uncertainty was assumed moderate for concentrations measured above these limit values (<30%), while the uncertainty was considered high for concentrations below them (>30%). Such an approach has already been adopted by (Ruel et al. 2011), which stated that the uncertainty on the micropollutant concentration in wastewater is generally comprised between 30 and 100% when the value measured is lower than 2.5 to 10 times the LQ, depending on the compound, and lower than 30% when higher than this value. Considering that, the removal of micropollutants was estimated only in several situations to limit the propagation of

207	uncertainties and provide more reliable results. When the compound was quantified above 5 times the LQ in influents
208	and above the LQ in effluents, the removal was conventionally calculated. In addition, when the compound was
209	quantified above the LQ in influents (>LQ or >5 x LQ) and below the LQ in effluents, the removal was estimated using
210	LQ/2 instead of the effluent concentration. The removals were not calculated when concentrations of both influents and
211	effluents were measured between the LQ and 5 times the LQ.
212	
213	Finally, 4 classes of behaviors were defined regarding the obtained removals: very good (>80%), good (60-80%),
214	moderate (30-60%) and poor (<30%).
215	
216	
217	RESULTS AND DISCUSSION
218	
219	1. Quality of influents in nominal and degraded WWTP configuration
220	
221	1.1. Conventional wastewater parameters in influents
222	
223	As displayed in Table 3, the pilot influents are characterized by relatively low values of the conventional wastewater
224	parameters in nominal WWTP configuration, particularly DOC (5.6 \pm 0.9 mgC/L), UV-254 (0.110 \pm 0.013 cm ⁻¹), COD
225	and BOD ₅ (26 \pm 11 and 4.8 \pm 3.5 mgO ₂ /L), TKN (1.5 \pm 0.2 mgN/L), NH ₄ ⁺ (< 0.3 mgN/L), TSS (3 \pm 1 mg/L) and TP
226	(<0.3 mgP/L). Indeed, this WWTP achieves very high removals of TSS (98%), COD (92%), TN (76%) and TP (95%)
227	in nominal configuration (Mailler et al. 2013). In addition, a substantial fraction of the DOC (\approx 20%) is due to residual
228	methanol from the post denitrification step (SIAAP source), in contrary to DOC from the degraded WWTP
229	configuration (no denitrification).
230	In this study, the WWTP effluents are rather less concentrated (DOC, UV-254, etc.) than effluents from other studies
231	focusing on PAC adsorption from wastewater. Besides, effluents with average DOC values of 9.6 - 14.4 mgC/L and
232	UV-254 of 0.239 - 0.397 cm ⁻¹ were recently studied by (Altmann et al. 2014), while DOC concentration of the water
233	studied by (Löwenberg et al. 2014) was 8.8 ± 1.2 mg/L. Similarly, (Margot et al. 2013) performed PAC adsorption on
234	biological treatment effluents with DOC of 7.3 \pm 1.9 mgC/L. Only the study of (Boehler et al. 2012) provides
235	micropollutant removal by PAC data with lower DOC concentration water, between 5.6 and 8.9 mgC/L.
236	

The degraded configuration, with partial nitrification and no denitrification, induces a notable increase of concentrations, particularly DOC ($7.5 \pm 0.5 \text{ mgC/L}$) and the UV-254 ($0.139 \pm 0.011 \text{ cm}^{-1}$). In addition, as no denitrification is performed in this WWTP configuration, the composition of the DOC is different without residue of methanol. The difference of DOC concentration is then even higher between the two WWTP configurations.

The concentration of NH_{4^+} in influents highlights the degradation of the nitrification step, with concentrations up to 4.4 $\pm 0.7 \text{ mgN/L}$. Similarly, the highest concentration of NO_3^- in influents from degraded configuration (23.7 $\pm 2.1 \text{ mgN/L}$) reflects the lack of denitrification. Finally, the TSS concentration is doubled when WWTP configuration is degraded (Table 3). Overall, the obtained quality of influents in this WWTP configuration is degraded and quite comparable to the one from (Margot et al. 2013).

The higher quantity of organic matter and most of pollutants in degraded WWTP configuration should be unfavorable to micropollutant adsorption. Indeed, it has been demonstrated that a higher DOC concentration induces a stronger competition effect (Delgado et al. 2012, Margot et al. 2013).

- 249
- 250

251 1.2. Micropollutant concentrations in influents from nominal WWTP configuration

252

Among the 54 PhPHs monitored, a total of 26 compounds were quantified at least during the nominal WWTP configuration. During this WWTP configuration, 14 were systematically quantified in influents, 12 in several samples but 28 were never quantified (supporting material - Table S4). The dissolved concentrations of the 26 compounds, in both nominal and degraded configuration of the WWTP, are presented in Table 2, with their occurrences.

257

8 compounds were measured in influents at average dissolved concentrations higher than 100 ng/L: sulfamethoxazole (993 ± 817 ng/L), ofloxacin (412 ± 315 ng/L), carbamazepine (215 ± 85 ng/L), atenolol (185 ± 51 ng/L), diclofenac (184 ± 91 ng/L), oxazepam (139 ± 128 ng/L) and erythromycin (124 ± 32 ng/L). 7 compounds were measured at concentrations lower than 10 ng/L, especially estrone (9 ± 3 ng/L) and ibuprofen (9 ± 5 ng/L). The 11 remaining compounds have average concentrations between 10 and 100 ng/L, i.e. naproxen (33 ± 28 ng/L), trimethoprim (64 ± 79 ng/L) or propranolol (97 ± 27 ng/L).

These concentrations are rather in accordance with data available in the literature for WWTP effluents (Deblonde et al.

265 2011, Loos et al. 2013, Luo et al. 2014, Miège et al. 2009), although they are overall in the lower part of the range (i.e.

analgesics). This could be due to i) the dilution of the Parisian wastewater (combined sewer) by parasite waters (Gasperi

et al. 2008) and ii) intense biological treatment including a total nitrification step, known to enhance the biodegradation
of micropollutants (Clara et al. 2005, McAdam et al. 2010). Sulfamethoxazole contributes about 30% of the total PhPHs
concentration, and its concentration lies in the upper part of the range found in the literature. No data were found for
testosterone, sulfadimerazine, sulfameter, and very scarce information are available for sulfadiazine, sulfathiazole,
fenofibrate and lorazepam. A short review as regard concentrations found in WWTP effluents for these compounds (23
references) is given in supporting material - Table S5.
In addition, most compounds are measured above 5 times their LQ (Table 3), leading to a moderate uncertainty on the

concentration. Sulfadimerazine, sulfadiazine, sulfameter, sulfathiazole, fenofibrate and testosterone are rather measured
between LQ and 5 times LQ, corresponding to high uncertainties, and the corresponding removals were consequently
not calculated.

- 277
- 278

279 1.3. Micropollutant concentrations in influents from degraded WWTP configuration

280

281 In the degraded configuration, 21 PhPHs were quantified. A similar pattern is observed but most of the PhPHs are 282 measured at higher average dissolved concentrations than in nominal WWTP configuration (Table 2), particularly 283 paracetamol (x200 based on average concentrations), ibuprofen (x100), ketoprofen (x10) and bezafibrate (x10). This is mainly explained by both seasonal variations (phase 1 June-October, phase 2 November-December) and the lack of 284 285 nitrification since this step has been identified as crucial for micropollutant biodegradation in biological treatments, especially for easily biodegradable molecules by biological treatments (Joss et al. 2005, Margot et al. 2013, Radjenović 286 et al. 2009). The degree of nitrification is then correlated to some PhPHs biodegradation (Margot et al. 2013), and a 287 total nitrification, as in nominal WWTP configuration, is known to be more efficient than a partial nitrification (Joss et 288 289 al. 2008), as in degraded WWTP configuration. This results in lower concentrations in discharges from nominal WWTP 290 configuration. Finally, the total concentration of the PhPHs doubles from 2 729 \pm 1 057 ng/L in nominal WWTP configuration to 4.956 ± 3.628 ng/L in degraded WWTP configuration, mainly due to the increase of paracetamol and 291 292 ibuprofen concentrations. In contrary, four compounds are measured at notably lower concentrations: sulfamethoxazole, 293 carbamazepine, ofloxacin and ciprofloxacin, probably because of seasonal variability.

294

A total of 59 other emerging micropollutants were also monitored during phase 2 in order to improve and enlarge the performance overview of the PAC treatment. Concentrations are presented in Table 3. Among them, monitored during degraded WWTP configuration, 34 were measured above the LQ. Only 3 pesticides (atrazine, diuron and isoproturon)
out the 23 pesticides investigated were quantified due to both a low occurrence in treated wastewater and high LQ.
Most of the compounds were always found in influents, especially PFOS, PFOA, bisphenol A, nonylphenols (NP),
para-tert-octylphenol (t-OP), bis(2-ethylhexyl) phthalate (DEHP), artificial sweeteners, triclosan and parabens.

301

302 A wide variety of dissolved concentrations is observed, from a few ng/L to more than 1 000 ng/L. Phthalate levels are rather high, ranging from 300 to 3 000 ng/L. Similarly, acesulfame and sucralose exhibit the highest concentrations by 303 304 far, respectively 8 725 \pm 602 and 7 150 \pm 745 ng/L, due to very poor removals by conventional WWTPs (Large et al. 305 2012). Artificial sweeteners have been recently recognized as a new class of emerging environmental contaminants (Lange et al. 2012), highly persistent, and their toxicity in the environment is still not well known. Bisphenol A, NP, 306 cyclamate and saccharin are found at lower concentrations (100-1 000 ng/L), as well as diuron, isoproturon, PFOS, 307 308 PFOA and parabens (10-100 ng/L). Then, all the PAHs present dissolved concentrations lower than 20 ng/L in the 309 effluents. Finally, as the concentrations presented for these micropollutants were acquired in degraded WWTP 310 configuration, it can be assumed that they should be lower or equal in nominal WWTP configuration. Levels found are very similar to other studies for pesticides, perfluorinated acids, bisphenol A, triclosan and phthalates (Bergé et al. 311 312 2013, Deblonde et al. 2011, Loos et al. 2013, Luo et al. 2014, Zareitalabad et al. 2013). In contrary, substantially higher 313 levels of sweeteners (Berset and Ochsenbein 2012, Ordóñez et al. 2012) and PAHs (Fatone et al. 2011, Qiao et al. 2014, Sánchez-Avila et al. 2011) were found in WWTP effluents in the literature. Similarly, levels of parabens found are 314 315 rather high compared to the literature (Chen et al. 2012, Yu et al. 2012b). In particular, concentrations are slightly 316 higher than those reported by (Geara-Matta 2012) for the same WWTP discharges, highlighting the impact of the 317 degraded configuration. More information about data available in the literature is given in supporting material - Table 318 S5.

Like PhPHs, most of these other emerging micropollutants feature concentrations higher than 5 times their LQ, except three PAHs (anthracene, benzo[a]anthracene and dibenzo[ah]anthracene) and some campaigns for NP, benzylbutyl phthalate (BBP), cyclamate and two parabens (benzyl and butyl paraben).

- 322
- 323

324 2. Efficacy of PAC in nominal and degraded WWTP configurations

325

326 2.1. Conventional wastewater parameter removal in nominal and degraded WWTP configurations

327

An overall improvement of the water quality is observed after the PAC treatment. In nominal WWTP configuration, 328 concentration removals of DOC ($35 \pm 24\%$), UV-254 ($32 \pm 14\%$), COD ($13 \pm 14\%$) and BOD₅ ($39 \pm 19\%$) are 329 330 observed (n=9, average PAC dose of 14 mg/L). These results are rather similar or lower than those of (Margot et al. 331 2013), despite the same average PAC dose. Actually, this difference could be due to the filtration unit after the PAC 332 contact tank in their study, which was recognized as the main cause of the general water quality improvements because it was biologically active. Furthermore, they evaluated between 20 and 35% the DOC removal due to the PAC against 333 $35 \pm 24\%$ and $23 \pm 4\%$ in nominal and degraded configurations respectively. Similarly, (Boehler et al. 2012) observed 334 a DOC removal between 15 and 48%, consistent with previously cited results. 335

336

In degraded WWTP configuration, the influent quality (Table 3) is degraded (Figure 4). Indeed, DOC is rather better removed both in percentage and concentration in nominal WWTP configuration, despite higher influent concentrations. In contrary, UV-254, COD, BOD₅ and TKN are similarly removed in both WWTP configurations. In addition, the presence of NH_4^+ in degraded WWTP effluents (Table 3) allows highlighting that a nitrification process occurs in the pilot, with a removal of about 20 ± 7 %. This confirms the findings of (Margot et al. 2013) even if authors quote a higher NH_4^+ removal ($85 \pm 20\%$) because of the biologically active filtration unit after the PAC contact tank. The formation of NO_3^- and NO_2^- confirms the nitrification process.

- 344
- 345

346 2.2. Micropollutant removal in nominal WWTP configuration

347

The overall performances of the pilot in both nominal and degraded WWTP configuration are presented in Figure 2. For each compound, the results from nominal configuration are given on the left hand and the results from degraded configuration on the right hand. A short literature review about the micropollutant removals by PAC is also provided in supporting material - Table S5.

352

In nominal WWTP configuration with a fresh PAC dose at 10 mg/L (n=3), paracetamol and ibuprofen are poorly removed, and 4 compounds are moderately removed by the pilot, including estrone (32-61%, min-max). In contrary, 9 substances are very well removed, including propranolol (96-98%), atenolol (86-92%), trimethoprim (84-98%), carbamazepine (86-97%), oxazepam (82-91%), bezafibrate (75-99%) and ciprofloxacin (76-91%). Finally, the 11 remaining compounds are well removed, i.e. ofloxacin (63-89%), naproxen (46-99%), sulfamethoxazole (53-72%) and diclofenac (72-85%).

359

360 To the best of our knowledge, papers dealing with the removal of micropollutants from wastewater by PAC are very 361 scarce, especially at large scale. Among the 26 PhPHs quantified, only 16 have been already studied in a comparable 362 context - adsorption from treated wastewater by PAC. Despite rather large standard deviations in our study, removals are similar to those of these studies (Altmann et al. 2014, Boehler et al. 2012, Löwenberg et al. 2014, Margot et al. 363 2013) for these 16 common PhPHs, as shown by supporting materials - Table S3 and Figure S3. Only ibuprofen and 364 365 estrone removals are notably lower in our study, probably because of their lower influent concentrations. In contrary, ciprofloxacin is better removed than in the literature. As the average fresh PAC doses are comparable for these studies 366 367 (14-15 mg/L), this suggests that micropollutant adsorption on PAC is likely to be a very stable phenomenon and that 368 these results could be extrapolated to other PAC treatments. A generalization of these results to other WWTPs and PAC processes could be then possible and relevant, even if efficacy in waters with slightly higher DOC concentrations 369 370 should be slightly lower. In addition, some papers have already observed comparable performances over PhPHs with 371 advanced oxidation processes such as ozonation (Altmann et al. 2014, Margot et al. 2013). For several compounds, 372 scarce data are available displaying their removal by grain activated carbon filter and/or PAC batch experiments with 373 surface water (paracetamol, roxithromycin, erythromycin and testosterone), while no reference was found for 374 sulfadimerazine, sulfameter, sulfadiazine, sulfathiazole, lorazepam and fenofibrate (supporting material - Table S5). 375 Overall, observed removals are similar to those from former studies (Altmann et al. 2014, Boehler et al. 2012, 376 Löwenberg et al. 2014, Margot et al. 2013, Ruel et al. 2012, Snyder et al. 2007, Westerhoff et al. 2005).

377

The removals vary due to variations of the fresh PAC doses (supporting material - Table S1) and of the influent concentrations. Despite this, the variations of removal percentages are relatively moderate for most of the compounds, around 5-15% between Q1 and Q3, except for paracetamol, sulfamethoxazole and bezafibrate, which have higher variations of 20-30%. Moreover, the observed variations are lower for well and very well removed compounds (i.e. beta blockers or trimethoprim), i.e. below 10%.

383

384

387 In degraded WWTP configuration, different impacts can be observed on PhPHs removal percentages, depending on the compound (Figure 4). First, paracetamol and ibuprofen are better removed in percentage in degraded configuration, 388 most likely due to their higher influent concentrations. Several compounds are rather similarly removed in both 389 390 configurations, i.e. both beta blockers, trimethoprim, sulfadiazine and ofloxacin, despite variations of influent 391 concentrations. Then, lower removals in percentage are observed for a majority of compounds, during degraded WWTP 392 configuration (Figure 4), despite higher influent concentrations. Nevertheless, the removed pollutant load is higher in degraded than in nominal WWTP configuration (ketoprofen, naproxen, diclofenac, trimethoprim, roxithromycin, 393 394 metronidazole, atenolol, propranolol, oxazepam, lorazepam, bezafibrate and estrone). In contrary, carbamazepine, 395 ciprofloxacin and sulfamethoxazole have lower removal in degraded configuration due to lower influent concentrations. This decrease of the micropollutant removals can be explained by both the variations of influent concentrations and the 396 397 water quality change. As demonstrated by higher DOC and UV-254, the influents in degraded WWTP exhibit higher 398 organic matter levels, inducing a stronger competition with micropollutants for adsorption on the PAC by direct site 399 competition and pore blocking (de Ridder et al. 2011, Delgado et al. 2012). In particular, the negative relation between 400 DOC concentration and micropollutant removals has already been highlighted in the literature (Altmann et al. 2014, 401 Margot et al. 2013).

Finally, the sum of PhPHs is similarly removed in nominal and degraded WWTP configurations (72-80% vs 70-81%).
This is mainly explained by higher removals of the high concentrated pollutants such as paracetamol and ibuprofen and
slightly lower removals of the other compounds.

405

406 Regarding the other emerging micropollutants monitored (Figure 3), atrazine (51-64%, min-max), isoproturon (51-83%) and to a greater extent diuron (82-96%) have a good affinity for PAC, confirming the suitability of the process for 407 408 pesticide removal. This confirms other studies observations (Margot et al. 2013, Ruel et al. 2012, Snyder et al. 2007), 409 even if atrazine is slightly less removed in our case (lower PAC dose). In addition, parabens (>70%), particularly ethyl 410 paraben (88-94%) and propyl paraben (91-96%), are also well or very well removed. As far as authors know, this is the first study displaying their fates within PAC in wastewater, as well as for phthalates and PAHs. In contrary, acesulfame 411 (9-19%), sucralose (6-26%), triclosan (18-29%) and PFOS (6-52%) are poorly removed by adsorption. For PFOA, no 412 clear conclusion can be drawn, since a systematic negative removal was observed, with concentrations always higher in 413 effluents than in influents. No sampling blanks were performed for the study, but this potential contamination may 414 result from Teflon[®] pipes or elements of the automatic samplers. 415

417 The 9 remaining compounds have variable removals but are rather moderately removed, such as saccharin (33-54%), DEHP (49-63%), bisphenol A (49-78%) or NP (3-97%). The removal of these substances from wastewater by 418 adsorption was poorly studied, but (Ruel et al. 2012) reported that within a granulated activated carbon treatment, 419 420 DEHP and NP are respectively moderately (30-70%) and poorly (<30%) removed. (Yu et al. 2008) also observed poor 421 removal for NP by activated carbon. Similarly, (Scheurer et al. 2010) observed that moderate to good removals can be 422 achieved by PAC adsorption for saccharin and sucralose, what is consistent with results obtained in our study. Finally, bisphenol A was found to be slightly better removed from wastewater by PAC in (Margot et al. 2013). Regarding 423 PAHs, the variation of removal is high but these compounds are overall not eliminated by the process or poorly 424 425 removed (10-40% for pyrene, fluoranthene, indeno[123]pyrene and benzo[ghi]perylene).

426

Even if the number of samples for these substances is limited, the variations of pesticide, sweetener and paraben removals from campaign to campaign are lower than 30%. Contrariwise, phthalates (except DEHP), PAHs, alkylphenols and perfluorinated acids have large variations of removals.

- 430
- 431

432 2.4. Water quality after the PAC treatment

433

The tertiary PAC treatment doesn't radically change the micropollutant pattern between influent and effluent (Table 3), 434 435 but it substantially decreases their dissolved concentrations. As the generalization of tertiary treatments to all WWTPs would depend strongly on regulations, having an idea of the levels that could be required for these compounds is 436 crucial. Instead of an overall removal (average removal of 80%), such as required in Switzerland, environmental quality 437 criteria (EQC) for freshwaters, comparable to environmental quality standards (EQS) for priority pollutants from (EC 438 2013), could be used as references. Such EQC have been proposed in Switzerland by the Ecotoxicity Centre of 439 440 EAWAG (supporting material - Table S6) (Götz et al. 2010, Kase et al. 2011). A comparison of the concentrations measured in effluents from the PAC treatment with these EQC or EQS enables to state if this process efficacy is 441 442 sufficient to reach references. However, these EQC/EQS apply for freshwaters and not effluents from WWTP, so a 443 dilution factor should be considered when evaluating the quality of the treatment and the compliance with regulations.

444

For PhPHs, naproxen, trimethoprim and atenolol concentrations are already measured below these proposed values before treatment. Ibuprofen, sulfamethoxazole, ciprofloxacin, propranolol, carbamazepine and bezafibrate are always measured below levels proposed after the treatment, while a fresh PAC dose of 20 mg/L allows decreasing diclofenac,
erythromycin and estrone concentrations below the EQC proposed in Switzerland. For diclofenac, this is particularly
interesting as this compound is now listed on the first watch list of the (EC 2013).

450 Among the other emerging micropollutants measured, some are listed as priority pollutants (EC 2013). Maximum 451 admissible concentration (MAC-EQS) and annual average environmental quality standards (AA-EQS) have been set up 452 for them in surface waters (supporting material - Table S5). All the compounds with MAC-EQS defined were always quantified below these values in the dissolved phase. For AA-EQS, different cases were encountered: pesticides, t-OP 453 454 and DEHP were always found below, NP was measured below only after treatment; PFOS and benzo[a] pyrene were still found above after treatment. Bisphenol A and triclosan are not included in (EC 2013), but have EOC proposed by 455 the Swiss Ecotoxicity Centre of EAWAG. Regarding it, bisphenol A was always measured below this criterion, even in 456 457 influents and before dilution, while triclosan was still measured above its limit value after treatment. However, for these compounds, the campaigns have been performed when the WWTP operated in degraded configuration, what should 458 lead to higher concentrations (Geara-Matta 2012). In addition, the WWTP discharges are diluted in the Seine River. 459

460

461

462 3. Influence of operation parameters and physico-chemical properties of the compounds

463

Different types of parameters can influence the adsorption processes (de Ridder et al. 2011, Delgado et al. 2012): structures and properties of the adsorbent, physico-chemical properties of the targeted compounds, water quality and composition and the operation parameters (dose of adsorbent, contact time, etc.). As explained previously, emerging micropollutant removals are lower when the WWTP operates in degraded configuration, because of stronger competition from effluent organic matter and higher concentrations. Results of this study also allow characterizing the relationship between fresh PAC dose and performances, as well as investigating the relations between physico-chemical properties of the compounds and their behaviors with PAC.

471

472 **3.1. Operation parameters**

473

474 The fresh PAC dose

To assess the influence of the fresh PAC dose, 4 doses (5-10-20-70 mg/L) were tested during nominal WWTP configuration and the results of the 12 PhPHs quantified in every campaigns of the phase 1 (Figure 4) were considered. Based on our results, the fresh PAC dose appears to be the leading operation parameter as regards its influence on performances. In particular, a significant positive correlation is found between the removal of the 26 PhPHs and the dose of PAC ($r_{spearman} = 0.962$; p-value < 0.001; $\alpha = 0.05$) considering the results in nominal WWTP configuration. Similarly, significant correlations are found between their individual removals and the PAC dose, as well as for DOC removal (supporting material - Table S6). In particular, diclofenac, sulfamethoxazole, metronidazole, sulfadiazine, atenolol, propranolol and carbamazepine have all spearman coefficient of correlation higher than 0.8 with p-value lower than 0.05.

As previously reported (Boehler et al. 2012, Snyder et al. 2007), the higher the PAC dose, the higher the removals. 484 Overall, the point of inflexion is reached around 10 mg/L for most of the compounds, explaining why the gain of 485 removal is relatively moderate between 10 and 20 mg/L and between 20 and 70 mg/L of PAC, despite the strong 486 487 correlation. In addition, while 9 compounds are very well removed at 10 mg/L of PAC, they are 14 at 20 mg/L and 3 at 488 5 mg/L (Figure 4). Similarly, the number of well or very well removed compounds increases from 12 at 5 mg/L of PAC, to 20 at 10 mg/L and 26 at 20 mg/L. Finally, applying a high dose of fresh PAC (70 mg/L) enables to achieve a 489 490 very high elimination of micropollutants, with removals higher than 90% for all compounds. In particular, diclofenac 491 which is on the first watch list of the (EC 2013), is moderately removed at 5 mg/L of fresh PAC, well removed at 10 492 mg/L and very well removed at 20 mg/L. Similarly, sulfamethoxazole is very impacted by the PAC dose. Finally, the 493 removal of the total PhPHs concentration is around 76% at 10 mg/L of PAC, 45% at 5 mg/L and 83% at 20 mg/L.

494

495 Total mass of the PAC in the bed

496 In normal operation of the pilot, the total mass of the PAC bed is close to 40 kg. Despite substantial variations of the total mass of PAC in the bed (supporting material - Table S1), no significant impact is observed on the performances, 497 498 highlighting that this parameter is minor. However, the presence of a high mass of PAC in the bed should slightly 499 enhance the performances compared to operating with only the fresh PAC dose, as it was observed with laboratory scale 500 experiments (not presented). This confirms that the observed removals on the pilot are mainly achieved by the fresh 501 PAC dose whereas the quantity of preloaded one in the bed allows a limited enhancement of the overall performances. 502 This observation has to be confirmed by laboratory scale tests in order to clearly identify the role of injected or 503 preloaded PAC.

506 The hydraulic velocity has theoretically an influence on contact time. Moreover, contact time has been identified as significantly impacting the adsorption process in the literature (Snyder et al. 2007). However, the hydraulic velocity/the 507 508 water flow has not a significant impact on the observed contact time in the CarboPlus® process because the variation of 509 the hydraulic velocity is coupled with a modification of the bed depth. Indeed, a higher velocity means a higher bed 510 depth by natural expansion, which balances the decrease of contact time. In addition, the experimental design doesn't 511 allow deeply assessing this point, contrary to the PAC dose. Nevertheless, although a slight impact is observed on efficacy for some compounds (supporting material - Figure S4) such as atenolol, diclofenac or oxazepam, it seems that 512 hydraulic velocity would not be a driving parameter of the process efficacy. 513

514

Finally, the solid retention time (SRT) of PAC and the presence of coagulant and flocculant were not studied in this paper. (Margot et al. 2013) have nevertheless observed a slight increase of micropollutant removal by PAC adsorption in presence of coagulant.

- 518
- 519

520 3.2. Influence of the physico-chemical properties of the compounds on their removals

521

522 The fate of many pollutants within PAC can be explained by their physico-chemical properties. First of all, the charge 523 of the compound is a crucial parameter in their removal. Indeed, all positively charged compounds (atenolol, 524 propranolol, trimethoprim, ciprofloxacin, norfloxacin) are well removed (>80%), regardless their other properties. This 525 has already been underlined in the literature (de Ridder et al. 2011, Margot et al. 2013). In fact, depending on the point 526 of zero charge (PZC) of the adsorbent, its surface can be neutral or slightly charged (positively or negatively) at influent pH (7-8). In addition, sorption of effluent organic matter, generally negatively charged in wastewater, on activated 527 528 carbon surface can switch (if initially neutral or positive) or increase (if already negative) the charge, resulting overall in 529 a surface negatively charged (Margot et al. 2013, Yu et al. 2012a). Therefore, this indicates that in this case the PAC 530 surface has negative charges inducing strong electrostatic attraction of positive compounds. In spite of their positive 531 charges, erythromycin and roxithromycin are less removed (50-60%), probably as a results of their high molecular masses (733.5 and 837.0 g/mol respectively; supporting material - Table S4), inducing a higher sensibility to 532 competition with organic matter and other compounds (Ji et al. 2010), and a size exclusion (Moreno-Castilla et al. 533 534 2003). For these compounds, a slight positive relation with hydrophobicity (log D_{OW}) is moreover observed, i.e. atenolol (80% - log D_{OW} = -1.99) vs. propranolol (94% - log D_{OW} = 0.98). The number of H-bond sites and the 535

536 compound structure do not seem to have any substantial influence for these compounds. Moreover, considering only 537 compounds with similar log D_{ow} and molecular mass, negatively charged compounds are less removed than positively 538 charged ones (i.e. trimethoprim, propranolol, naproxen, diclofenac, sulfamethoxazole).

539

540 For neutral and negatively charged compounds (Figure 2 and supporting material - Table S3), the hydrophobicity and 541 the structure of the compound seem to play a major role in removal. The influence of hydrophobicity, polarizability or structure of the compound (functional groups allowing H-bond or π - π binding) has already been reported (de Ridder et 542 al. 2010, Delgado et al. 2012). Indeed, the adsorption of the neutral compounds seems to be influenced by their 543 544 hydrophobicity, as reflected by a significant positive correlation ($r_{spearman} = 0.587$; p-value < 0.05; $\alpha = 0.05$) found between log Dow and removals of PhPHs. (Westerhoff et al. 2005) observed a similar trend. In addition to the 545 546 hydrophobicity, specific interactions between PhPHs functional groups and PAC should be involved. Indeed, (de Ridder 547 et al. 2010) have reported that H-binding and π - π interactions become very important at low hydrophobicity, inducing 548 possible substantial differences in the fate of compounds with similar hydrophobicity.

549

550 No significant or specific relations were found for negative compounds between removal and properties. 551 Hydrophobicity was not identified as enhancing their adsorption, in contrary to what (Margot et al. 2013) observed. 552 Absence of direct relationship between adsorption and hydrophobicity was also observed for some antibiotics by (Ji et 553 al. 2010). These compounds are more likely influenced by their structures and their fates should be driven by a balance 554 between electrostatic repulsion and specific interactions with PAC surface. (Moreno-Castilla et al. 2003) concluded that 555 aromatic compounds are mainly physisorbed on activated carbon, confirming the prevalence of these weak interactions. 556 Some functional groups may enhance adsorption such as aromaticity and N-heterocycles (Delgado et al. 2012), explaining the high removal of ofloxacin although this compound is negatively charged. For instance, ofloxacin and 557 diclofenac have three heterocycles, known to enhance adsorption on activated carbon (Delgado et al. 2012), while 558 559 sulfamethoxazole and sulfadiazine have only one heterocycle.

560

561

562 CONCLUSIONS

563

Among the solutions to reduce the emerging contaminant discharges into the environment, the implementation of a tertiary treatment in actual WWTP is more and more considered. Besides, emerging micropollutants, particularly pharmaceuticals, may be included in modifications of existing regulations. For instance, diclofenac, $17-\alpha$ - and $17-\beta$ estradiols are now on the first watch list of the (EC 2013), and the Swiss Centre for applied ecotoxicology has proposed environmental quality criteria for several emerging micropollutants.

569

In this context, the fate of 113 micropollutants within a large scale PAC adsorption pilot (CarboPlus[®]) has been studied. 54 pharmaceuticals and hormones (PhPHs) and 59 other micropollutants, such as pesticides, personal care products, phthalates, PAHs or artificial sweeteners, were then monitored. The 14 campaigns performed allowed assessing the overall performances of the process for these compounds, and the influence of the WWTP configuration, the operation parameters and the physico-chemical properties of the micropollutants on the process efficacy.

575

26 PhPHs were quantified in influents from nominal WWTP configuration including 8 substances with concentrations higher than 100 ng/L. Sulfamethoxazole is predominant with the highest average concentration, accounting for about 30% in average of the sum of the 26 PhPHs. In contrary, the remaining molecules were measured whether at low (<10 ng/L), like estrone or lorazepam, or at intermediate concentrations (10-100 ng/L), like naproxen or trimethoprim. 6 of them are poorly or moderately removed by the process (<60%), i.e. paracetamol, ibuprofen or sulfamethoxazole, and 11 are rather well removed (60-80%), especially diclofenac, naproxen or oxazepam. The 9 remaining substances are very well eliminated (>80%), i.e. beta blockers, carbamazepine or trimethoprim

The change of the WWTP configuration has a substantial impact on the influents quality (DOC, UV-254, BOD₅, nitrogen species, TSS) and most of the PhPHs have higher concentrations in this configuration, highlighting both the seasonal variations and the removal performed due to denitrification step in nominal configuration. Paracetamol and ibuprofen concentrations are multiplied by 100. PhPHs are overall slightly less removed in percentage in this WWTP configuration but a higher flux is eliminated. This is consistent considering the higher concentrations and the degradation of the influents quality which probably enhances the competition effects.

Concentrations of the other micropollutants during degraded WWTP configuration are variable but phthalates and artificial sweeteners are present at very high concentrations, up to 1 000-10 000 ng/L, bisphenol A and NP between 100 and 1 000 ng/L, and pesticides, perfluorinated acids, PAHs and parabens below 100 ng/L. Except parabens and pesticides, which are largely removed (50-95%), most of the other micropollutants are poorly to moderately removed. However, similarly to PhPHs, it can be assumed that performances would be higher in nominal WWTP configuration for these compounds.

595 The fresh PAC dose has been identified as the main operation parameter which significantly influences the

performances for micropollutant removals. The total mass of PAC in the bed was not identified as influencing the performances in the studied range of mass. Overall, in nominal WWTP configuration, the CarboPlus[®] process reduces in average the sum of the PhPHs of about 53% at a dose of PAC of 5 mg/L, 76% at 10 mg/L and more than 83% at 20 mg/L. Finally, a high dose of PAC allows to achieve very high removals (>90%) of all the PhPHs, but at higher costs.

The molecular charge seems to be the most important property influencing the fate of micropollutants. However, the size of the molecule can be a limiting factor because high molecular weight compounds are more sensitive to organic matter competition. A higher dose is then needed for heavy compounds compared to others, at comparable other molecular properties. For neutral or negative compounds, hydrophobicity and structure of the molecule, particularly the presence of specific functional groups, become very important in their fates.

605

This pilot is still operating in 2014, and a new type of activated carbon is tested. This micro-grain activated carbon (μ GAC) has an intermediary size between PAC (<50 μ m) and grain (>1 mm), that facilitates the bed handling (no need of chemical addition) together with good performances for micropollutants elimination. Moreover, μ GAC selected is produced from regenerated activated carbon and is regenerated several times, ensuring a reduction of cost and a higher sustainability of the process.

611

612 ACKNOWLEDGEMENT

613

This study has been performed within the framework of the OPUR research program. The authors would like to thank the SIAAP (Céline Briand and Julien Pouillaude), the LEESU (Damien Lherm), the ISA (Audrey Bulete, Antoine Vauchez, Mikaël Tournier and Loic Spinner) and the SAUR (Séverine Bareilles) teams for their technical support and their active participation to the sampling campaigns. Similarly, authors thank the technical teams of the Technological University of Compiègne (UTC) and the Water Technology Centre (TZW) for their participation in analyses.

619 **REFERENCES**

620

621 Abegglen, C. and Siegrist, H. (2012) Micropolluants dans les eaux résiduaires 622 urbaines. Etapes de traitement supplémentaire dans les stations d'épuration. (in 623 French). Environment, S.F.O.f.t. (ed), p. 87. 624 Altmann, J., Ruhl, A.S., Zietzschmann, F. and Jekel, M. (2014) Direct comparison 625 of ozonation and adsorption onto powdered activated carbon for micropollutant 626 removal in advanced wastewater treatment. Water Research 55(0), 185-193. 627 Barrek, S., Cren-Olivé, C., Wiest, L., Baudot, R., Arnaudguilhem, C. and Grenier-Loustalot, M.-F. (2009) Multi-residue analysis and ultra-trace 628 629 quantification of 36 priority substances from the European Water Framework Directive by GC-MS and LC-FLD-MS/MS in surface waters. Talanta 79(3), 712-722. Bergé, A., Cladière, M., Gasperi, J., Coursimault, A., Tassin, B. and Moilleron, R. (2013) Meta-analysis of environmental contamination by phthalates. 630 631 632 633 Environmental Science and Pollution Research, 1-20. Bergé, A., Gasperi, J., Rocher, V., Gras, L., Coursimault, A. and Moilleron, R. (2014) Phthalates and alkylphenols in industrial and domestic effluents: Case of Paris conurbation (France). Science of the Total Environment 488-489(0), 26-35. 634 635 636 637 Berset, J.-D. and Ochsenbein, N. (2012) Stability considerations of aspartame in 638 the direct analysis of artificial sweeteners in water samples using high-639 performance liquid chromatography-tandem mass spectrometry (HPLC-MS/MS). 640 Chemosphere 88(5), 563-569. Boehler, M., Zwickenpflug, B., Hollender, J., Ternes, T., Joss, A. and Siegrist, 641 642 H. (2012) Removal of micropollutants in municipal wastewater treatment plants by 643 powder-activated carbon, International Water Association, London, Royaume-Uni. 644 Bolong, N., Ismail, A.F., Salim, M.R. and Matsuura, T. (2009) A review of the effects of emerging contaminants in wastewater and options for their removal. 645 Desalination 239(1-3), 229-246. 646 647 Bressy, A., Gromaire, M.C., Lorgeoux, C., Saad, M., Leroy, F. and Chebbo, G. (2012) Towards the determination of an optimal scale for stormwater quality 648 649 management: Micropollutants in a small residential catchment. Water Research 650 46(20), 6799-6810. 651 Cecen, F. and Aktas, Ö. (2012) Activated carbon for water and wastewater treatment integration of adsorption and biological treatment, Wiley-VCH, 652 653 Weinheim, Germany. Chen, Z.-F., Ying, G.-G., Lai, H.-J., Chen, F., Su, H.-C., Liu, Y.-S., Peng, F.-Q. and Zhao, J.-L. (2012) Determination of biocides in different environmental matrices by use of ultra-high-performance liquid chromatography-tandem mass 654 655 656 spectrometry. Analytical and Bioanalytical Chemistry 404(10), 3175-3188. Clara, M., Strenn, B., Gans, O., Martinez, E., Kreuzinger, N. and Kroiss, H. (2005) Removal of selected pharmaceuticals, fragrances and endocrine disrupting 657 658 659 660 compounds in a membrane bioreactor and conventional wastewater treatment plants. 661 Water Research 39(19), 4797-4807. 662 Clara, M., Scharf, S., Scheffknecht, C. and Gans, O. (2007) Occurrence of 663 selected surfactants in untreated and treated sewage. Water Research 41, 4339-664 4348. Daughton, C.G. and Ternes, T.A. (1999) Pharmaceuticals and personal care 665 666 products in the environment: Agents of subtle change? Environmental Health 667 Perspectives 107, 907-938. 668 de Ridder, D.J., Villacorte, L., Verliefde, A.R.D., Verberk, J.Q.J.C., Heijman, S.G.J., Amy, G.L. and van Dijk, J.C. (2010) Modeling equilibrium adsorption of 669 670 organic micropollutants onto activated carbon. Water Research 44(10), 3077-3086. 671 de Ridder, D.J., Verliefde, A.R., Heijman, S.G., Verberk, J.Q., Rietveld, L.C., van der Aa, L.T., Amy, G.L. and van Dijk, J.C. (2011) Influence of natural organic matter on equilibrium adsorption of neutral and charged pharmaceuticals 672 673 onto activated carbon. Water Sci Technol 63(3), 416-423. 674 675 de Ridder, D.J., Verliefde, A.R.D., Schoutteten, K., van der Linden, B., 676 Heijman, S.G.J., Beurroies, I., Denoyel, R., Amy, G.L. and van Dijk, J.C. (2013) 677 Relation between interfacial energy and adsorption of organic micropollutants 678 onto activated carbon. Carbon $53(\overline{0})$, 153-160. 679 Deblonde, T., Cossu-Leguille, C. and Hartemann, P. (2011) Emerging pollutants in 680 wastewater: A review of the literature. International Journal of Hygiene and Environmental Health 214(6), 442-448. Delgado, L.F., Charles, P., Glucina, K. and Morlay, C. (2012) The removal of 681 682 endocrine disrupting compounds, pharmaceutically activated compounds and 683 684 cyanobacterial toxins during drinking water preparation using activated carbon-A review. Science of the Total Environment 435-436(0), 509-525. Ebie, K., Li, F., Azuma, Y., Yuasa, A. and Hagishita, T. (2001) Pore distribution effect of activated carbon in adsorbing organic micropollutants 685 686 687

688 from natural water. Water Research 35(1), 167-179. EC (2013) Directive 2013/39/EU of the European Parliament amending Directives 689 690 2000/60/EC and 2008/105/EC as regards priority substances in the field of water 691 policy. JO-EU L226/1. Fatone, F., Di Fabio, S., Bolzonella, D. and Cecchi, F. (2011) Fate of aromatic 692 693 hydrocarbons in Italian municipal wastewater systems: An overview of wastewater treatment using conventional activated-sludge processes (CASP) and membrane 694 695 bioreactors (MBRs). Water Research 45(1), 93-104. Gasperi, J., Kafi-Benyahia, M., Lorgeoux, C., Moilleron, R., Gromaire, M.C. and 696 697 Chebbo, G. (2008) Wastewater quality and pollutant loads in combined sewers 698 during dry weather periods. Urban Water Journal 5(4), 305-314. Gasperi, J., Geara, D., Lorgeoux, C., Bressy, A., Zedek, S., Rocher, V., El Samrani, A., Chebbo, G. and Moilleron, R. (2014) First assessment of triclosan, triclocarban and paraben mass loads at a very large regional scale: Case of 699 700 701 702 Paris conurbation (France). Science of the Total Environment 493(0), 854-861. 703 Geara-Matta, D. (2012) Flux and sources of parabens, triclosan and triclocarban 704 in dense urban areas: comparison between Paris and Beyrouth (in French), Ecole 705 des Ponts ParisTech. 706 Götz, C., Stamm, C., Fenner, K., Singer, H., Schärer, M. and Hollender, J. 707 (2010) Targeting aquatic microcontaminants for monitoring: exposure categorization and application to the Swiss situation. Environmental Science and 708 709 Pollution Research 17(2), 341-354. 710 Halling-Sørensen, B., Nors Nielsen, S., Lanzky, P.F., Ingerslev, F., Holten Lützhøft, H.C. and Jørgensen, S.E. (1998) Occurrence, fate and effects of 711 712 pharmaceutical substances in the environment- A review. Chemosphere 36(2), 357-713 393 Heberer, T. (2002) Occurrence, fate, and removal of pharmaceutical residues in the aquatic environment: a review of recent research data. Toxicology Letters 714 715 716 131(1-2), 5-17. 717 Ji, L., Liu, F., Xu, Z., Zheng, S. and Zhu, D. (2010) Adsorption of 718 Pharmaceutical Antibiotics on Template-Synthesized Ordered Micro- and Mesoporous Carbons. Environmental Science & Technology 44(8), 3116-3122. Jones, O.A.H., Voulvoulis, N. and Lester, J.N. (2001) Human Pharmaceuticals in 719 720 721 the Aquatic Environment a Review. Environmental Technology 22(12), 1383-1394. 722 Joss, A., Keller, E., Alder, A.C., Göbel, A., McArdell, C.S., Ternes, T. and 723 Siegrist, H. (2005) Removal of pharmaceuticals and fragrances in biological 724 wastewater treatment. Water Research 39(14), 3139-3152. 725 Joss, A., Siegrist, H. and Ternes, T.A. (2008) Are we about to upgrade 726 wastewater treatment for removing organic micropollutants? Water Sci Technol 57(2), 251-255. 727 728 Kase, R., Eggen, R.I.L., Junghans, M., Götz, C. and Hollender, J. (2011) Waste Water - Evaluation and Management. Einschlag, P.F.S.G. (ed), InTech. 729 730 Lange, F., Scheurer, M. and Brauch, H.-J. (2012) Artificial sweeteners-a 731 recently recognized class of emerging environmental contaminants: a review. Analytical and Bioanalytical Chemistry 403(9), 2503-2518. Loos, R., Carvalho, R., António, D.C., Comero, S., Locoro, G., Tavazzi, S., 732 733 734 Paracchini, B., Ghiani, M., Lettieri, T., Blaha, L., Jarosova, B., Voorspoels, 735 S., Servaes, K., Haglund, P., Fick, J., Lindberg, R.H., Schwesig, D. and Gawlik, B.M. (2013) EU-wide monitoring survey on emerging polar organic contaminants in wastewater treatment plant effluents. Water Research 47(17), 6475-6487. 736 737 Löwenberg, J., Zenker, A., Baggenstos, M., Koch, G., Kazner, C. and Wintgens, T. (2014) Comparison of two PAC/UF processes for the removal of micropollutants 738 739 740 from wastewater treatment plant effluent: Process performance and removal 741 efficiency. Water Research 56(0), 26-36. 742 Luo, Y., Guo, W., Ngo, H.H., Nghiem, L.D., Hai, F.I., Zhang, J., Liang, S. and 743 Wang, X.C. (2014) A review on the occurrence of micropollutants in the aquatic 744 environment and their fate and removal during wastewater treatment. Science of the Total Environment 473-474(0), 619-641. Mailler, R., Gasperi, J., Rocher, V., Gilbert-Pawlik, S., Geara-Matta, D., Moilleron, R. and Chebbo, G. (2013) Biofiltration vs conventional activated 745 746 747 sludge plants: what about priority and emerging pollutants removal? 748 749 Environmental Science and Pollution Research, 1-12. 750 Margot, J., Kienle, C., Magnet, A., Weil, M., Rossi, L., de Alencastro, L.F., 751 Abegglen, C., Thonney, D., Chèvre, N., Schärer, M. and Barry, D.A. (2013) 752 Treatment of micropollutants in municipal wastewater: Ozone or powdered activated carbon? Science of the Total Environment 461-462(0), 480-498. 753 McAdam, E.J., Bagnall, J.P., Koh, Y.K.K., Chiu, T.Y., Pollard, S., Scrimshaw, M.D., Lester, J.N. and Cartmell, E. (2010) Removal of steroid estrogens in 754 755 756 carbonaceous and nitrifying activated sludge processes. Chemosphere 81(1), 1-6.

757 Miège, C., Choubert, J.M., Ribeiro, L., Eusèbe, M. and Coquery, M. (2009) Fate 758 of pharmaceuticals and personal care products in wastewater treatment plants -759 Conception of a database and first results. Environmental Pollution 157(5), 760 1721-1726. 761 Moreno-Castilla, C., Bautista-Toledo, I., Ferro-Garcia, M.A. and Rivera-Utrilla, 762 J. (2003) Influence of support surface properties on activity of bacteria 763 immobilised on activated carbons for water denitrification. Carbon 41, 1743-764 1749. 765 Ordóñez, E.Y., Quintana, J.B., Rodil, R. and Cela, R. (2012) Determination of 766 artificial sweeteners in water samples by solid-phase extraction and liquid 767 chromatography-tandem mass spectrometry. Journal of Chromatography A 1256(0), 768 197 - 205. 769 Qiao, M., Qi, W., Liu, H. and Qu, J. (2014) Occurrence, behavior and removal of 770 typical substituted and parent polycyclic aromatic hydrocarbons in a biological 771 wastewater treatment plant. Water Research 52(0), 11-19. Radjenović, J., Petrović, M. and Barceló, D. (2009) Fate and distribution of 772 773 pharmaceuticals in wastewater and sewage sludge of the conventional activated 774 sludge (CAS) and advanced membrane bioreactor (MBR) treatment. Water Research 43(3), 831-841. 775 776 Rocher, V., Paffoni, C., Goncalves, A., Guerin, S., Azimi, S., Gasperi, J., 777 Moilleron, R. and Pauss, A. (2012) Municipal wastewater treatment by 778 biofiltration: comparisons of various treatment layouts. Part 1: assessment of 779 carbon and nitrogen removal. Water Science and Technology 65(9), 1705-1712. Ruel, S.M., Choubert, J.M., Esperanza, M., Miege, C., Navalon Madrigal, P., 780 Budzinski, H., Le Menach, K., Lazarova, V. and Coquery, M. (2011) On-site evaluation of the removal of 100 micro-pollutants through advanced wastewater 781 782 783 treatment processes for reuse applications. Water Science and Technology 63(11), 2486-2497. 784 785 Ruel, S.M., Choubert, J.M., Budzinski, H., Miege, C., Esperanza, M. and Coquery, 786 M. (2012) Occurrence and fate of relevant substances in wastewater treatment 787 plants regarding Water Framework Directive and future legislations. Water 788 Science and Technology 65(7), 1179-1189. 789 Sánchez-Avila, J., Fernandez-Sanjuan, M., Vicente, J. and Lacorte, S. (2011) 790 Development of a multi-residue method for the determination of organic 791 micropollutants in water, sediment and mussels using gas chromatography-tandem 792 mass spectrometry. Journal of Chromatography A 1218(38), 6799-6811. 793 Scheurer, M., Brauch, H.-J. and Lange, F. (2009) Analysis and occurrence of 794 seven artificial sweeteners in German waste water and surface water and in soil 795 aquifer treatment (SAT). Analytical and Bioanalytical Chemistry 394(6), 1585-796 1594. 797 Scheurer, M., Storck, F.R., Brauch, H.-J. and Lange, F.T. (2010) Performance of 798 conventional multi-barrier drinking water treatment plants for the removal of 799 four artificial sweeteners. Water Research 44(12), 3573-3584. 800 Snyder, S.A., Adham, S., Redding, A.M., Cannon, F.S., DeCarolis, J., Oppenheimer, J., Wert, E.C. and Yoon, Y. (2007) Role of membranes and activated 801 802 carbon in the removal of endocrine disruptors and pharmaceuticals. Desalination 803 202(1-3), 156-181. 804 Vulliet, E., Cren-Olivé, C. and Grenier-Loustalot, M.-F. (2011) Occurrence of 805 pharmaceuticals and hormones in drinking water treated from surface waters. Environmental Chemistry Letters 9(1), 103-114. Westerhoff, P., Yoon, Y., Snyder, S. and Wert, E. (2005) Fate of Endocrine-806 807 808 Disruptor, Pharmaceutical, and Personal Care Product Chemicals during Simulated 809 Drinking Water Treatment Processes. Environmental Science & Technology 39(17), 810 6649-6663. 811 Yu, J., Lv, L., Lan, P., Zhang, S., Pan, B. and Zhang, W. (2012a) Effect of effluent organic matter on the adsorption of perfluorinated compounds onto 812 813 activated carbon. Journal of Hazardous Materials 225-226(0), 99-106. 814 Yu, K., Li, B. and Zhang, T. (2012b) Direct rapid analysis of multiple PPCPs in municipal wastewater using ultrahigh performance liquid chromatography-tandem mass spectrometry without SPE pre-concentration. Analytica Chimica Acta 738(0), 815 816 817 59-68. 818 Yu, Z., Peldszus, S. and Huck, P.M. (2008) Adsorption characteristics of 819 selected pharmaceuticals and an endocrine disrupting compound-Naproxen, 820 carbamazepine and nonylphenol-on activated carbon. Water Research 42(12), 2873-821 2882. 822 Zareitalabad, P., Siemens, J., Hamer, M. and Amelung, W. (2013) 823 Perfluorooctanoic acid (PFOA) and perfluorooctanesulfonic acid (PFOS) in surface waters, sediments, soils and wastewater - A review on concentrations and 824

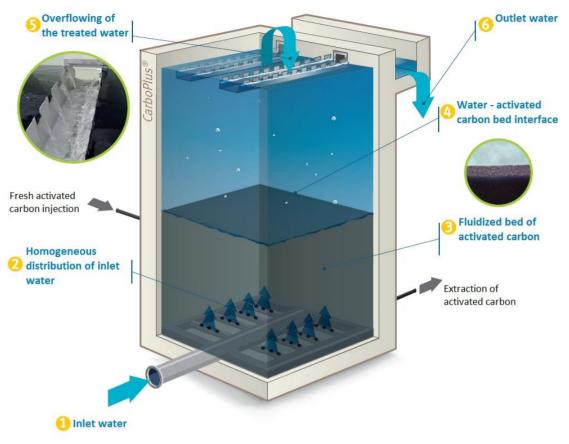


Figure 1 - Layout of the 20 m³ (5 m x 4 m²) CarboPlus® pilot (SAUR source)

Table 1 - Operation parameters for the 14 campaigns performed

	Date	WWTP configuration	PAC dose	Hydraulic velocity	Total PAC mass (kg)
C1*	02/07/13		70		
C2	09/07/13		20	8	
C3	16/07/13		5		
C4	23/07/13	Nominal	5	10	
C5	30/07/13	Nominal	20		-
C6	06/08/13		10	6	
C7	20/08/13		10		
C8	27/08/13		5	8	
C9	04/11/13		10	7	62
C10	19/11/13				46
C11	27/11/13				52
C12	03/12/13	Degraded	10	7	53
C13	10/12/13	_			45
C14	17/12/13				58

834 835

838

	Ta	ble 2 - P	ollutants studied and t	heir analytical methods	
Groups	N ^a	n ^b	Reference	Extraction ^c	Analysis ^d
Antibiotics	31	14			
Analgesic	5	14		SPE	
Beta blockers	2	14	(V_{1})	<i>Autotrace</i> ®	I C MEME
Amiglation	4	14	(Vulliet et al. 2011)	Consta Volton Onsta III Do	LC-MSMS

(Barrek et al. 2009)

(Vulliet et al. 2011)

(Barrek et al. 2009)

(Vulliet et al. 2011)

(Bressy et al. 2012)

(Bergé et al. 2014)

(Bergé et al. 2014)

(Gasperi et al. 2014)

(Scheurer et al. 2009)

StrataX® or Oasis HLB®

cartridges

SPE

Autotrace®

StrataX® cartridges

SPE

Manual

Oasis HLB® cartridges

SPE

Autotrace®

Bakerbond SDB1 cartridges

Total micropollutants 113

Anxiolytics

Hypolipemiants

Hormones

Chlorinated solvents

Perfluorinated acids

Pesticides / herbicides

/ insecticides

Bisphenol A

PAHs ^f

Alkylphenols

Phthalates

PCPs ^f

Sweeteners

4

2

10

3

2

23

1

13

2

4

7

4

14

14

14

5

5

5

5

3

5

5

3

4

^a N = number of substances.

^b n = number of campaigns performed.

^c SPE = solid phase extraction.

^d Analytical methods: LC = liquid chromatography, GC = gas chromatography, GC-MS = GC with mass spectrometry, LC-MSMS

= LC with tandem mass spectrometry.

^e LQ = limit of quantification.

^f PAHs = polycyclic aromatic hydrocarbons; PCPs = personal care products.

839 840

LQ^e

0.2 - 110

0.04 - 1.0

0.02

1.0 - 2.0

0.7 - 2.4

1.0

0.2 - 23

1.0

6 - 177

0.2

10

0.2

0.6 - 100

100

2.8 - 10

50 - 250

GC-MS

LC-MSMS

GC-MS

LC-MSMS

LC-MSMS

GC-MS

LC-MSMS

LC-MSMS

8	4	1
~		~

			Nominal WWTI (n=		guration		Degraded WWT (n=		Iguration
	LQ		Influent		Effluent		Influent		Effluent
	(ng/L)		Concentration		Concentration		Concentration		Concentratio
	(116/12)	Ν	(ng/L)	Ν	(ng/L)	Ν	(ng/L)	Ν	(ng/L)
		11	$Av. \pm SD$	1	$Av. \pm SD$	11	$Av. \pm SD$	11	$Av. \pm SD$
			(min - max)		(min - max)		(min - max)		(min - max)
			Convention	al wast	ewater parameters	5			
	0.01	0.10	0.110 ± 0.013	0.10	0.075 ± 0.020		0.139 ± 0.011		0.109 ± 0.00
UV 254 nm (cm ⁻¹)	0.01	9/9	(0.087 - 0.120)	9/9	(0.047 - 0.106)	5/5	(0.126 - 0.152)	5/5	(0.099 - 0.12
2001 07		0.10	5.6 ± 0.9	0.10	3.5 ± 1.2		7.5 ± 0.5		5.8 ± 0.4
DOC (mgC/L)	0.5	9/9	(4.4 - 7.0)	9/9	(1.2 - 5.2)	5/5	(7.1 - 8.3)	5/5	(5.4 - 6.4)
			26 ± 11		34 ± 22		26 ± 2		21 ± 2
COD (mgO ₂ /L)	4	9/9	(17 - 47)	9/9	(13 - 74)	5/5	(23 - 28)	5/5	(19 - 25)
			4.8 ± 3.3		3.5 ± 1.9		5.8 ± 0.9		3.0 ± 0.5
BOD ₅ (mgO ₂ /L)	0.5	9/9	(2.8 - 11.0)	9/9	(1.3 - 7.0)	5/5	(4.9 - 7.0)	5/5	(2.4 - 3.7)
							4.4 ± 0.7		3.5 ± 0.5
NH4 ⁺ (mgN/L)	0.3	0/9	<lq< td=""><td>0/9</td><td><lq< td=""><td>5/5</td><td>(3.2 - 5.0)</td><td>5/5</td><td>(2.9 - 4.0)</td></lq<></td></lq<>	0/9	<lq< td=""><td>5/5</td><td>(3.2 - 5.0)</td><td>5/5</td><td>(2.9 - 4.0)</td></lq<>	5/5	(3.2 - 5.0)	5/5	(2.9 - 4.0)
			0.21 ± 0.17		0.15 ± 0.16		(5.2 ± 5.0) 0.7 ± 0.02		(2.9 ± 4.0) 0.29 ± 0.16
NO_2^- (mgN/L)	0.02	9/9	(0.06 - 0.23)	8/9	(< LQ - 0.30)	5/5	(0.05 - 0.09)	5/5	0.29 ± 0.10 (0.03 - 0.46
NO ₃ ⁻ (mgN/L)	0.4	9/9	9.1 ± 2.9	9/9	8.5 ± 2.8	5/5	23.7 ± 2.1	5/5	24.4 ± 2.5
			(5.6 - 11.7)		(4.9 - 13.7)		(20.6 - 25.4)		(20.7 - 26.6
TKN (mgN/L)	0.5	9/9	1.5 ± 0.2	9/9	1.2 ± 0.1	5/5	4.9 ± 0.5	5/5	3.9 ± 0.4
			(1.1 - 1.8)		(0.9 - 1.3)		(4.1 - 5.5)		(3.4 - 4.4)
PO_4^{3-} (mgP/L)	0.1	0/5	<lq< td=""><td>0/5</td><td><lq< td=""><td>0/5</td><td><lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<></td></lq<></td></lq<>	0/5	<lq< td=""><td>0/5</td><td><lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<></td></lq<>	0/5	<lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<>	0/5	<lq< td=""></lq<>
TP (mgP/L)	0.3	0/5	<lq< td=""><td>0/5</td><td><lq< td=""><td>0/5</td><td><lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<></td></lq<></td></lq<>	0/5	<lq< td=""><td>0/5</td><td><lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<></td></lq<>	0/5	<lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<>	0/5	<lq< td=""></lq<>
	2.0	0./0	3 ± 1	0.10	9 ± 7	- (-	6 ± 2	- 1-	5 ± 2
TSS (mg/L)	2.0	9/9	(2 - 5)	9/9	(2 - 21)	5/5	(4 - 9)	5/5	(2 - 6)
	•			ceutical	s and hormones		, , , , , , , , , , , , , , , , , , , ,		
		0.10	34 ± 18		11 ± 10		367 ± 90	- /-	145 ± 28
Ketoprofen	0.3	9/9	(13 - 61)	6/9	(< LQ - 30)	5/5	(273 - 501)	5/5	(109 - 182
			(13 + 31) 33 ± 28		12 ± 13		154 ± 39	-	52 ± 9
Naproxen	0.7	8/8	(8 - 83)	5/8	(< LQ - 34)	5/5	(122 - 220)	5/5	(43 - 64)
			$(3 \div 33)$ 33 ± 22		(<lq -="" 34)<br="">32 ± 18</lq>		(122 - 220) 5 870 ± 2 597		1030 ± 922
Paracetamol	0.04	6/9	35 ± 22 (<lo -="" 70)<="" td=""><td>6/9</td><td>32 ± 18 (<<i>LQ</i> - 56)</td><td>5/5</td><td>$(3\ 610\ -\ 10\ 350)$</td><td>5/5</td><td>1030 ± 92 (349 - 2650</td></lo>	6/9	32 ± 18 (< <i>LQ</i> - 56)	5/5	$(3\ 610\ -\ 10\ 350)$	5/5	1030 ± 92 (349 - 2650
			~ ~ /						
Ibuprofen	1.0	2/9	9	5/9	9 ± 5	5/5	951 ± 360	5/5	432 ± 200
			(<lq -="" 13)<="" td=""><td></td><td>(<lq -="" 14)<="" td=""><td></td><td>(590 - 1 439)</td><td></td><td>(220 - 699</td></lq></td></lq>		(<lq -="" 14)<="" td=""><td></td><td>(590 - 1 439)</td><td></td><td>(220 - 699</td></lq>		(590 - 1 439)		(220 - 699
Diclofenac	0.3	9/9	184 ± 91	9/9	52 ± 51	5/5	384 ± 76	5/5	171 ± 19
Dicioicitae	0.5)/)	(95 - 309))()	(3 - 166)	5/5	(301 - 508)	5/5	(147 - 196)
Sulfamethoxazole	1.0	0/0	993 ± 817	9/9	419 ± 318	5/5	233 ± 179	2/5	130
Sulfamethoxazole	1.0	9/9	(175 - 3 010)	9/9	(37 - 798)	5/5	(70 - 470)	3/5	(<lq -="" 235<="" td=""></lq>
0.7	10	0.10	412 ± 314	0.10	70 ± 74		39 ± 7	2/5	18
Ofloxacin	10	9/9	(14 - 911)	8/9	(<lq -="" 218)<="" td=""><td>5/5</td><td>(34 - 51)</td><td>3/5</td><td>(<lq -="" 24)<="" td=""></lq></td></lq>	5/5	(34 - 51)	3/5	(<lq -="" 24)<="" td=""></lq>
			175 ± 93		22 ± 17		13 ± 3		5
Ciprofloxacin	1.0	8/9	(64 - 312)	6/9	(< LQ - 51)	5/5	(11 - 17)	3/5	(<lq -="" 8)<="" td=""></lq>
			64 ± 79		$\frac{4\pm 2}{4\pm 2}$		43 ± 28	-	6 ± 3
Trimethoprim	0.3	9/9	(8 - 222)	8/9	(< LQ - 8)	5/5	(9 - 74)	4/5	(< LQ - 9)
			(3 - 222) 19 ± 3		$(\langle LQ - \delta \rangle)$ 8 ± 3		(9-74) 30 ± 4		(< LQ - 9) 25 ± 4
Metronidazole	0.2	9/9		8/9		5/5		5/5	
			(15 - 24)	-	(<lq -="" 12)<="" td=""><td></td><td>(23 - 33)</td><td></td><td>(18 - 29)</td></lq>		(23 - 33)		(18 - 29)
Roxithromycin	1.0	4/4	99 ± 53	4/4	35 ± 33	5/5	175 ± 59	4/5	99 ± 42
			(57 - 173)		(5 - 76)		(126 - 271)		(<lq -="" 155<="" td=""></lq>
Norfloxacin	1.0	5/9	80 ± 36	3/9	17	0/5	<lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<>	0/5	<lq< td=""></lq<>
1 (officiation	1.0	517	(<lq -="" 118)<="" td=""><td>517</td><td>(<lq -="" 29)<="" td=""><td>0/5</td><td><u>ب</u>بر</td><td>0/5</td><td><u>`</u></td></lq></td></lq>	517	(<lq -="" 29)<="" td=""><td>0/5</td><td><u>ب</u>بر</td><td>0/5</td><td><u>`</u></td></lq>	0/5	<u>ب</u> بر	0/5	<u>`</u>
Erythromycin	1.0	4/4	124 ± 32	4/4	50 ± 38	0/5	<lq< td=""><td>0/5</td><td>4.0</td></lq<>	0/5	4.0
Eryunomycin	1.0	4/4	(97 - 170)	4/4	(23 - 106)	0/5	<lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<>	0/5	<lq< td=""></lq<>
G 16 11 ·		e 15	4			C / -	• •	e /=	
Sulfadimerazine	1.0	2/9	(< <i>LQ</i> - 7)	1/9	(<lq -="" 4)<="" td=""><td>0/5</td><td><lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<></td></lq>	0/5	<lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<>	0/5	<lq< td=""></lq<>
	1		10 ± 6	1	4 ± 3		4 ± 1	-	1 ± 1
Sulfadiazine	1.0	9/9	(1 - 21)	7/9	(< LQ - 8)	5/5	(2 - 5)	4/5	(< LQ - 2)
	1.0	1/9	(1 - 21) (<lq -="" 4)<="" td=""><td>0/9</td><td></td><td>0/5</td><td>(2 - 3) <lq< td=""><td>0/5</td><td></td></lq<></td></lq>	0/9		0/5	(2 - 3) <lq< td=""><td>0/5</td><td></td></lq<>	0/5	
Sulfamator		1/9	$(\leq LU - 4)$	0/9	<lq< td=""><td>0/5</td><td>∣ <lų< td=""><td>0/5</td><td><lq< td=""></lq<></td></lų<></td></lq<>	0/5	∣ <lų< td=""><td>0/5</td><td><lq< td=""></lq<></td></lų<>	0/5	<lq< td=""></lq<>
Sulfameter Sulfathiazole	1.0	3/9	1	0/9	<lq< td=""><td>0/5</td><td><lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<></td></lq<>	0/5	<lq< td=""><td>0/5</td><td><lq< td=""></lq<></td></lq<>	0/5	<lq< td=""></lq<>

Table 3 - Micropollutant and conventional wastewater parameter concentrations in influents and effluents during nominal and degraded WWTP configuration

			(1 - 2)						
Atenolol	0.02	9/9	185 ± 51	9/9	28 ± 24	5/5	588 ± 156	5/5	176 ± 42
Propranolol	0.02	9/9	(124 - 251) 97 ± 27	9/9	(2 - 67) 5 ± 5	5/5	$\frac{(454 - 858)}{197 \pm 46}$	5/5	(126 - 236) 14 ± 5
			(66 - 131) 215 ± 85		(1 - 15) 41 ± 43		(162 - 276) 30 ± 2		(10 - 23) 8 ± 2
Carbamazepine	1.0	9/9	(19 - 321) 139 ± 128	8/9	(< <i>LQ</i> - 113) 29 ± 24	5/5	(27 - 32) 409 ± 132	4/5	(< <i>LQ</i> - <i>11</i>) 176 ± 31
Oxazepam	1.2	8/8	(20 - 354)	8/8	29 ± 24 (5 - 65)	5/5	(251 - 615)	5/5	(145 - 210)
Lorazepam	1.9	3/8	11 (< <i>LQ</i> - 27)	1/8	(<lq -="" 6)<="" td=""><td>4/5</td><td>34 ± 2 (<<i>LQ</i> - 36)</td><td>5/5</td><td>18 ± 3 (15 - 23)</td></lq>	4/5	34 ± 2 (< <i>LQ</i> - 36)	5/5	18 ± 3 (15 - 23)
Bezafibrate	0.7	8/8	36 ± 42 (1 - 102)	6/8	8 ± 9 (< <i>LQ</i> - 26)	5/5	369 ± 330 (151 - 940)	5/5	187 ± 149 (84 - 447)
Fenofibrate	2.4	1/8	(< LQ - 4)	0/8	<lq< td=""><td>0/8</td><td><lq< td=""><td>0/8</td><td><lq< td=""></lq<></td></lq<></td></lq<>	0/8	<lq< td=""><td>0/8</td><td><lq< td=""></lq<></td></lq<>	0/8	<lq< td=""></lq<>
Estrone	1.0	4/8	7 ± 2 (< <i>LQ</i> - 10)	5/8	9 ± 10 (< <i>LQ</i> - 26)	5/5	12 ± 3 (9 - 15)	5/5	5 ± 1 (4 - 7)
Testosterone	1.0	1/8	(<lq -="" 1)<="" td=""><td>0/8</td><td><lq< td=""><td>2/5</td><td>(<lq -="" 1)<="" td=""><td>0/5</td><td><lq< td=""></lq<></td></lq></td></lq<></td></lq>	0/8	<lq< td=""><td>2/5</td><td>(<lq -="" 1)<="" td=""><td>0/5</td><td><lq< td=""></lq<></td></lq></td></lq<>	2/5	(<lq -="" 1)<="" td=""><td>0/5</td><td><lq< td=""></lq<></td></lq>	0/5	<lq< td=""></lq<>
ΣPhPHs	-	-	2729 ± 1057 (1358 - 5158)	-	752 ± 517 (75 - 1 593)	-	9 892 ± 2 569 (6 844 - 13 910)	-	2625 ± 1007 (1711 - 4226)
		1	• • •	erging	micropollutants		(0007 10 910)		(1 / 11 / 220)
Atrazine	0.2					5/5	4 ± 1	5/5	2 ± 1
Auazine	0.2					5/5	(3 - 5) 25 ± 5	5/5	(1 - 2) 4 ± 1
Diuron	0.2					5/5	(19 - 30)	5/5	(2 - 5)
Isoproturon	0.2					5/5	34 ± 22 (11 - 62)	5/5	12 ± 6 (5 - 20)
PFOA	1.0					5/5	25 ± 9 (14 - 36)	5/5	37 ± 19 (16 - 57)
PFOS	1.0					5/5	44 ± 12 (24 - 56)	5/5	31 ± 13 (14 - 47)
Bisphenol A	10					5/5	$\frac{(21-50)}{259 \pm 148}$ (142 - 513)	5/5	78 ± 24 (52 - 112)
NP	100					5/5	$\frac{(142 - 513)}{841 \pm 681}$ $(143 - 1.895)$	3/5	(32 - 112) 286 (<lq -="" 390)<="" td=""></lq>
t-OP	0.6	-				5/5	17 ± 5 (10 - 22)	5/5	16 ± 15 (4 - 39)
DEHP	100					4/4	$\frac{1413 \pm 862}{(919 - 2704)}$	4/4	861 ± 489 (377 - 1 300)
DEP	100					5/5	$991 \pm 443 \\ (441 - 1 \ 644)$	5/5	437 ± 149 (298 - 646)
DnBP	100					5/5	$\frac{(111 + 1011)}{932 \pm 631}$ $(355 - 1986)$	5/5	$\frac{(290 - 010)}{607 \pm 383}$ $(309 - 1.280)$
Acesulfame	50					4/4	$\frac{(333 \pm 700)}{8725 \pm 602}$ (7900 - 9300)	4/4	$\frac{(50)}{7525 \pm 665}$ (6700 - 8200)
Cyclamate	50					2/4	$\frac{(1 + 260)^{-1}}{240}$ (< <i>LQ</i> - 430)	0/4	<lq< td=""></lq<>
Saccharin	50					4/4	$\frac{1}{1} \frac{355 \pm 572}{(840 - 2\ 100)}$	4/4	703 ± 284 (450 - 1 100)
Sucralose	250					4/4	$7\ 150 \pm 545 \\ (6\ 500 - 7\ 800)$	4/4	5875 ± 222 (5600 - 6100)
Triclosan	10					3/3	135 (<i>121 - 158</i>)	3/3	103 (86 - 119)
Methyl paraben	2.8					3/3	65 (31 - 102)	3/3	11 (10 - 13)
Ethyl paraben	3.5					3/3	40 (27 - 57)	2/3	5 (<lq -="" 7)<="" td=""></lq>
Propyl paraben	3.3					3/3	56 (37 - 81)	2/3	5 (<lq -="" 5)<="" td=""></lq>
Benzyl paraben	3.0]				1/3	(<lq -="" 4)<="" td=""><td>1/3</td><td>$(\langle LQ - 3 \rangle)$</td></lq>	1/3	$(\langle LQ - 3 \rangle)$
Butyl paraben	2.8					2/3	10 (<lq -="" 13)<="" td=""><td>0/3</td><td><lq< td=""></lq<></td></lq>	0/3	<lq< td=""></lq<>
Fluorene	0.2					3/3	1.7 (0.3 - 4.2)	3/3	3.0 (2.6 - 3.4)
Phenanthrene	0.2	J				3/3	16.7	3/3	24.2

			(13.0 - 19.1)	
Anthracene	0.2	3/3	0.6 (0.5 - 0.8)	3/3
Fluoranthene	0.2	3/3	3.2 (2.6 - 3.9)	3/3
Pyrene	0.2	3/3	2.4 (2.1 - 2.8)	3/3
Benzo[a]anthracene	0.2	0/3	<lq< td=""><td>1/3</td></lq<>	1/3
Chrysene	0.2	3/3	1.5 (1.2 - 1.6)	3/3
Benzo[b]fluoranthene	0.2	3/3	3.3 (2.2 - 4.0)	3/3
Benzo[k]fluoranthene	0.2	3/3	1.2 (0.9 - 1.4)	3/3
Benzo[a]pyrene	0.2	3/3	1.4 (0.9 - 1.7)	3/3
Indeno[123]pyrene	0.2	3/3	2.0 (1.0 - 2.6)	3/3
Dibenzo[ah]anthracene	0.2	3/3	0.5 (0.3 - 0.6)	3/3
Benzo[ghi]perylene	0.2	3/3	2.0 (1.1 - 2.6)	3/3

 $\begin{array}{l} LQ = limit \ of \ quantification. \\ N = number \ of \ quantification \ / \ campaigns \ performed. \\ Av. \pm SD = average \ \pm \ standard \ deviation \ (calculated \ only \ when \ N > 3). \end{array}$

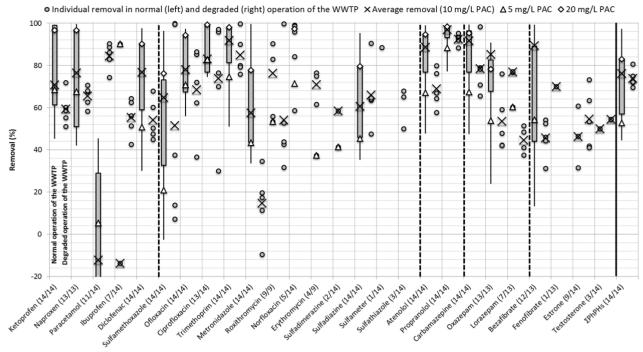
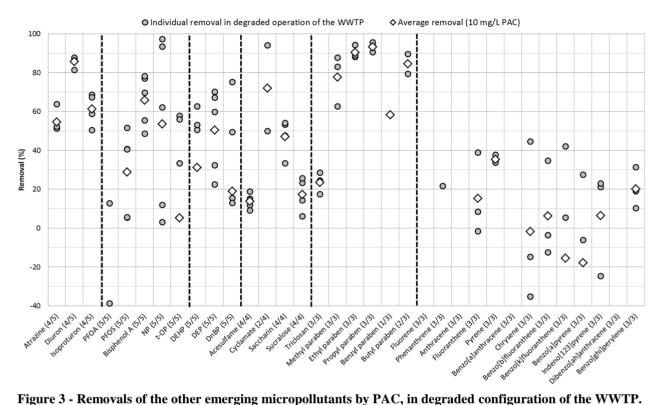
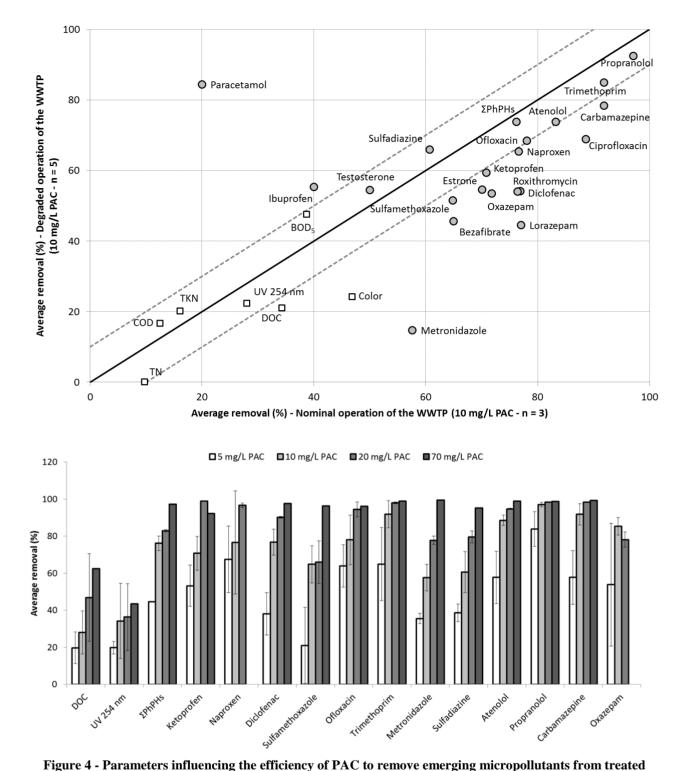




Figure 2 - Removals of pharmaceuticals and hormones (N=26) by PAC treatment during nominal (left) and degraded (right) WWTP configurations (min - Q1 - Q2 - Q3 - max or individual values) - occurrence in brackets 846



849 Figure 3 - Removals of the other emerging micropollutants by PAC, in degraded configuration of the WWTP.



wastewater - influence of the water quality (top) and the fresh PAC dose (bottom)