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Mathieu Cladière, Céline Bonhomme, Lauriane Vilmin, Johnny Gasperi, Nicolas Flipo, et al.. Modelling the fate of nonylphenolic compounds in the Seine River – part 1: Determination of in-situ attenuation rate constants. *Science of the Total Environment*, 2014, 468 -469, pp.1050-1058. 10.1016/j.scitotenv.2013.09.028 . hal-00870525

**HAL Id: hal-00870525**

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Submitted on 15 Oct 2013

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# 1 Modelling the fate of nonylphenolic compounds in 2 the Seine River - part 1: determination of *in-situ* 3 attenuation rate constants

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8 DOI: [10.1016/j.scitotenv.2013.09.028](https://doi.org/10.1016/j.scitotenv.2013.09.028)

## 9 Abstract

10 Assessing the fate of endocrine disrupting compounds (EDC) in the environment is currently a key issue for  
11 determining their impacts on aquatic ecosystems. The 4-nonylphenol (4-NP) is a well known EDC and results  
12 from the biodegradation of surfactant nonylphenol ethoxylates (NPnEO). Fate mechanisms of NPnEO are well  
13 documented but their rate constants have been mainly determined through laboratory experiments.

14 This study aims at evaluating the *in-situ* fate of 4-NP, nonylphenol monoethoxylate (NP<sub>1</sub>EO) and nonylphenolic  
15 acetic acid (NP<sub>1</sub>EC). Two sampling campaigns were carried out on the Seine River in July and September 2011,  
16 along a 28 km-transect downstream Paris City. The field measurements are used for the calibration of a sub-  
17 model of NPnEO fate, included into a hydro-ecological model of the Seine River (ProSe). The timing of the  
18 sampling is based on the Seine River velocity in order to follow a volume of water. Based on our results, *in-situ*  
19 attenuation rate constants of 4-NP, NP<sub>1</sub>EO and NP<sub>1</sub>EC for both campaigns are evaluated. These rate constants  
20 vary greatly. Although the attenuation rate constants in July are especially high (higher than 1 d<sup>-1</sup>), those  
21 obtained in September are lower and consistent with the literature. This is probably due to the biogeochemical  
22 conditions in the Seine River. Indeed, the July sampling campaign took place at the end of an algal bloom  
23 leading to an unusual bacterial biomass while the September campaign was carried out during common  
24 biogeochemical status. Finally, the uncertainties on measurements and on the calibration parameters are  
25 estimated through a sensitivity analysis.

26 This study provides relevant information regarding the fate of biodegradable pollutants in an aquatic  
27 environment by coupling field measurements and a biogeochemical model. Such data may be very helpful in the  
28 future to better understand the fate of nonylphenolic compounds or any other pollutants at the basin scale.  
29

## 30 *Keywords:*

31 *Nonylphenol and nonylphenol ethoxylates, attenuation rate constant, endocrine disrupting compounds*  
32 *modelling, surface water*

# 1. Introduction

Over the last 20 years, the scientific community paid a special attention to endocrine disrupting compounds (EDC) due to their toxicity on aquatic wildlife (Petrovic *et al.*, 2004). Among these compounds, the nonylphenol ethoxylates (NPnEO), and especially the 4-nonylphenol (4-NP), are of prime interest since concentrations of several hundred nanogrammes per liter were measured in all environmental and urban waters (Giger *et al.*, 1984; Ahel *et al.*, 1994). 4-NP is more estrogenic and more toxic than NPnEO (Soto *et al.*, 1991). Jugan *et al.* (2009) and Fenet *et al.* (2003) have reported the role played by the 4-NP in the estrogenic activity occurring in both surface water and sediment compartments. Due to its toxicity, the 4-NP have been included in the list of the 33 priority pollutants in the European water framework Directive 2000/60/EC (European Commission, 2000). More recently the European Directive 2008/105/EC established the environmental quality standard for 4-NP at 300 ng/L in surface water (European Commission, 2008). The 4-NP mainly originates from the biodegradation of NPnEO (readily biodegradable compounds) which are used as non-ionic surfactants in many industrial and domestic applications. The worldwide production of NPnEO reached 500,000 tons in 2000 and is decreasing because of regulations (Ying *et al.*, 2002). The biodegradation pathways of NPnEO are currently well known (John and White, 1998; Jonkers *et al.*, 2001; Giger *et al.*, 2009). Basically, the NPnEO can be biodegraded into 4-NP through an oxidative pathway leading to nonylphenolic acids (e.g. nonylphenol acetic acid: NP<sub>1</sub>EC) as biodegradation intermediates, or through a non-oxidative pathway leading to short chain nonylphenol ethoxylates (e.g. nonylphenol mono ethoxylates: NP<sub>1</sub>EO) as intermediates (Giger *et al.*, 2009). Finally, the 4-NP can be mineralized under well oxygenated conditions (Gabriel *et al.*, 2005). The biodegradation rate constants of NPnEO have been mostly determined through bioreactor laboratory experiments (Staples *et al.*, 2001; Jurado *et al.*, 2009; Karahan *et al.*, 2010). However, such experiments fail to represent the complexity of freshwater ecosystems, including the spatial and temporal heterogeneity and the numerous biological, physical and chemical parameters which may interfere with the dynamics of the NPnEO degradation. Jurado *et al.*, (2009) used NPnEO mixture as the sole source of carbon for microorganisms and found biodegradation rate constants of NPnEO about 0.2 h<sup>-1</sup>, while Staples *et al.*, (2001) used synthetic river water to perform a river die-away experiment (closer to *in-situ* conditions) and found biodegradation rate constants ranging from 0.04 d<sup>-1</sup> to 0.10 d<sup>-1</sup> for NP<sub>9</sub>EO and from 0.08 to 0.09 d<sup>-1</sup> for 4-NP. The difficulty to assess the NPnEO concentrations and their biodegradation relies in the fact that there is a lack of commercial pure reference standards which can be used for a reliable

1 analysis, and that various commercial mixtures exist with different chain lengths according to their use  
2 (Ayorinde *et al.*, 1999).

3 To date, only one study focuses on the determination of *in-situ* attenuation rate constants of NPnEO and 4-NP  
4 based on a hydrodynamic and biogeochemical model (Jonkers *et al.*, 2005). However, these attenuation rate  
5 constants have to be confirmed since:

- 6 i) *In-situ* biodegradation was determined in estuarine saline water. Authors suggest that biodegradation  
7 may significantly change in freshwater;
- 8 ii) The knowledge of biodegradation pathways has been updated; NP<sub>1</sub>EC is now recognized as a  
9 biodegradation precursor of 4-NP (Montgomery-Brown *et al.*, 2008; Giger *et al.*, 2009).

10 Thus, it is crucial to assess the *in-situ* biodegradation of short chain nonylphenol and 4-NP in surface water in  
11 order to evaluate their environmental fate and their impact on aquatic wildlife surrounding heavily urbanized  
12 area such as the Seine River downstream of Paris, France.

13 A global project on modelling NPnEO was launched in 2009. The first part aims at evaluating the attenuation  
14 rate constants; the second aims at modelling the fate of nonylphenolic compounds at the annual scale in order to  
15 validate the parameters calibrated in the first part and forecast future profiles of the Seine River. This study deals  
16 with the first part of the NPnEO modelling project, while the second part is handled in a companion paper with  
17 the simulation of annual time series of NPnEO and 4-NP concentrations for a reference year (2010) in the Seine  
18 River. At last, a forecast of nonylphenolic compound concentrations in the Seine River is attempted for the 21<sup>st</sup>  
19 century according to global changes scenarios (Cladière *et al.*, 2013a).

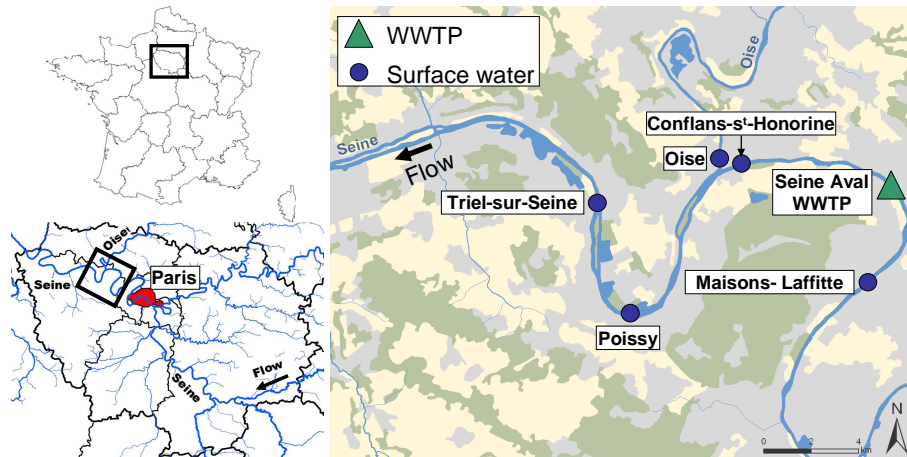
20 Therefore the goal of this first part is to determine the *in-situ* attenuation rate constants of 4-NP, NP<sub>1</sub>EC and  
21 NP<sub>1</sub>EO in the Seine River. To achieve this goal, two sampling campaigns are carried out and the data are used to  
22 calibrate a hydro-ecological model implemented for the Seine River. A special attention has been paid to the  
23 small scale spatial and temporal variabilities of the concentrations which are firstly *in-situ* assessed and then  
24 used in the model calibration procedure. Finally, the sensitivity analysis of the model to the biodegradation  
25 parameters is performed based on an approach “one factor at a time” (OFAT).

26 The calibrated parameters are then validated in the companion paper according to 11 monthly sampling  
27 campaigns carried out in 2010 at 3 sites on the Seine River and the Oise River as well as the effluent of the  
28 largest WWTP of the Parisian Metropolitan Area (Cladière *et al.* 2013a).

## 2. Materials and methods

### 2.1. Study site

This study focuses on the Seine River, downstream of Paris City (annual average flow in Paris =  $300 \text{ m}^3/\text{s}$ ,  $120 \text{ m}^3/\text{s}$  in summer). The investigated transect of the Seine River is 28 km long from Maisons-Laffite to Triel-sur-Seine (Fig.1).



**Fig. 1** Investigated transect of the Seine River (28 km-long) downstream of Paris and sampling sites

6  
7  
8

Four sampling sites are considered along this transect (Maisons-Laffite, Conflans-s<sup>l</sup>-Honorine, Poissy and Triel-sur-Seine) and two for the effluents of Seine Aval Waste Water Treatment Plant (WWTP) and the Oise River (Fig.1).

This transect is selected based on two major criteria. First, there are only two significant tributaries in the Seine River between Maisons-Laffite and Triel-sur-Seine, i.e. the effluents of Seine Aval WWTP and the Oise River (Fig.1). The Seine Aval WWTP is the biggest treatment plant of the Parisian Metropolitan Area and treats approximately  $1,666,000 \text{ m}^3$  per day (70 % of Parisian wastewater) with an average effluent flow of  $19 \text{ m}^3/\text{s}$ . The Oise River is one of the largest tributaries of the Seine River with an average discharge flow of  $95 \text{ m}^3/\text{s}$  ( $30 \text{ m}^3/\text{s}$  in summer). These two inflows account for 30 % to 40 % of the Seine River flow at Triel-sur-Seine during low-flow conditions. The small number of lateral inflows along this transect simplifies the determination of the boundary conditions for the modelling procedure. The second criterion is the distance between Maisons-Laffite and Triel-Sur-Seine. According to the average Seine River velocity ( $\approx 0.15 \text{ m/s}$ ) during low-flow conditions ( $< 150 \text{ m}^3/\text{s}$ ), the transit time along this transect is approximately 60 h. This duration is similar to the half-lives of 4-NP and NPnEO found in the literature: between 2 and 7 days (Staples et al., 2001; Jonkers et al., 2005).

22

## 2.2. Sampling campaigns

Two sampling campaigns were carried out in July and September 2011. These campaigns are designed to acquire data for the calibration of the ProSe model and the assessment of attenuation rate constants of nonylphenolic compounds in the Seine River. The timing of the sampling is estimated according to the Seine River flow velocity in order to collect samples in the same water volume from Maisons-Laffitte to Triel-sur-Seine (Lagrangian approach). The Seine River velocity is simulated using the hydrodynamic module of the ProSe model (see section 2.3), with measured upstream boundary conditions (national discharge gauging station located in Paris City). The description of the sampling campaigns (kilometre points, hours of sampling and flows) are provided in Table 1. The sampling campaigns were exclusively carried out under dry weather conditions in order to ensure that the Seine Aval WWTP and the Oise River are the sole lateral inflows (no wet weather sources). Due to an unexpected rain event, the July campaign was stopped at Poissy (45 h).

**Table 1** Sampling campaign description for July 2011 and September 2011.

Sampling points	Kilometre point (km)	July 2011 (Tw = 21°C)			September 2011 (Tw = 19°C)		
		Hours of sampling		Flow (m <sup>3</sup> /s)	Hours of sampling		Flow (m <sup>3</sup> /s)
Maisons-Laffitte	715.4	10/07	8:42	72	29/09	8:00	97
Seine Aval (WWTP)	720.1	10/07	19:24	16	29/09	15:30	18
Conflans-s <sup>t</sup> -Honorine	728.2	11/07	16:30	88	30/09	7:00	115
Oise River	728.7	11/07	17:06	32	30/09	7:30	28
Poissy	734.9	12/07	5:35	120	30/09	19:00	143
Triel-sur-Seine	743.6	*		-	01/10	18 :50	143
Total sampling time		45 h			59 h		

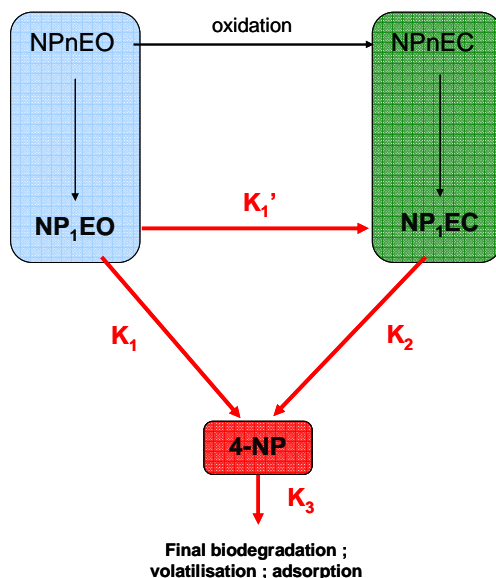
\*: The July sampling campaign was stopped at Poissy due to a rain event. Tw = water temperature

Samples are collected from bridges in the middle of the river with 2 L glass bottles. The analytical protocol is described in details by Cladière *et al.*, (2013b). Briefly, the water samples are filtered through a 0.45 µm porosity glass fiber filter (GF/F, Whatmann). Only the dissolved phase is kept and analyzed as the suspended particles account approximately for 5 % of total concentrations of NP<sub>1</sub>EO, NP<sub>1</sub>EC and 4-NP within the Seine River (Cladière *et al.*, 2010). After a spiking with a surrogate mixture (octylphenol-d17, NP<sub>1</sub>EO-d2), 250 mL of dissolved phase is extracted by solid phase extraction on OASIS<sup>®</sup> HLB cartridges (200 mg/6 mL; Waters<sup>®</sup>). Finally, the extracts are analyzed by means of an ultra-performance liquid chromatography coupled to a tandem mass spectrometer (UPLC-MS-MS; Waters<sup>®</sup>) using an internal standard mixture (linear compounds: 4-nNP, nNP<sub>1</sub>EO and nNP<sub>2</sub>EC). Based on this analytical protocol, the concentrations of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO are quantified for surface water and WWTP effluents. Given that no pure standard is commercially available for NP<sub>3</sub>EO to NP<sub>15</sub>EO, only semi-quantitative analyses are performed for these compounds. A semi-quantitative

1 analysis does not provide the real value of compound concentration but a concentration equivalent value  
 2 proportional to the real concentration (the factor is not known because of the lack of pure standard). For each  $n$   
 3 (from 3 to 15), dividing the NP $n$ EO peak area by the peak area of the relative internal standard nNP $_1$ EO give the  
 4 concentration equivalent of the NP $n$ EO. These concentration equivalent values enable comparisons from one site  
 5 to another but not between compounds (e.g. the factor of proportionality of NP $_3$ EO is different from that of  
 6 NP $_{15}$ EO).

### 7 2.3. Modelling tool: the ProSe model

8 The hydrodynamic and biogeochemical model ProSe was firstly developed to simulate the impacts of human  
 9 activities on nitrogenous and phosphorous pollutions in the Seine River and its tributaries (Even *et al.*, 1998;  
 10 Flipo *et al.*, 2007; Even *et al.*, 2007). All mathematical equations used by the ProSe model to simulate the river  
 11 flowing, pollutant advection and other biogeochemical parameters are provided by Even *et al.* (1998). For this  
 12 study the biogeochemical module is updated to take into account the biodegradation pathways of NP $n$ EO  
 13 following Giger *et al.*, (2009) scheme (Fig. 2).



14  
 15 **Fig. 2** Fate schema of nonylphenolic compounds in a well oxygenated surface water. The attenuation rate  
 16 constant  $K_1$ ,  $K_1'$ ,  $K_2$  and  $K_3$  are determined by the ProSe model. The biodegradation of NP $n$ EO and  
 17 NP $n$ EC is introduced in the ProSe model as precursor inputs of NP $_1$ EO and NP $_1$ EC along the Seine River.  
 18

19 In the Figure 2, the attenuation rate constants  $K_1$ ,  $K_1'$  and  $K_2$  only represent the biodegradation of NP $_1$ EO or  
 20 NP $_1$ EC into their products (NP $_1$ EC or 4-NP). On the contrary,  $K_3$  represent the global attenuation of 4-NP due to  
 21 biodegradation, volatilisation and adsorption onto particles.

1 Assuming first-order kinetics (Jonkers *et al.*, 2003; Jurado *et al.*, 2009), the attenuation rate constants  $K_1$ ,  $K_1'$ ,  $K_2$   
2 and  $K_3$  are calibrated using a trial-error procedure to optimise the matching of the modelled longitudinal profiles  
3 of 4-NP,  $NP_1EO$  and  $NP_1EC$  with the measured ones. Jonkers *et al.* (2005) suggest, during the sensitivity  
4 analysis of their model, that the sorption onto particles have a very limited influence on the fate of nonylphenolic  
5 compounds in estuarine water. Indeed, set sorption to 0 or multiply their optimum value by 5 lead to small  
6 variations of  $NP_1EO$  and  $NP_1EC$  dissolved concentrations. In addition, Cladiere *et al.* (2010) showed that the  
7 particulate concentrations of nonylphenolic compounds account for less than 5% of the total concentration.  
8 Therefore, the sorption onto particles of  $NP_1EO$  and  $NP_1EC$  has been considered as negligible. Similarly to  
9 Jonkers *et al.* (2005), the water-air exchange of the nonylphenolic compounds (except for 4-NP) has been  
10 considered as insignificant due to very low Henry's constants (e.g.  $NP_1EO$  and  $NP_2EO \approx 0.0003 \text{ Pa}\cdot\text{m}^3/\text{mole}$ ).  
11 Consequently, three equations are implemented in the biogeochemical module of the ProSe model:

$$\frac{d[NP_1EO]}{dt} = -K_1 [NP_1EO] - K_1' [NP_1EO] + \text{precursor inputs} \quad \text{Equation 1}$$

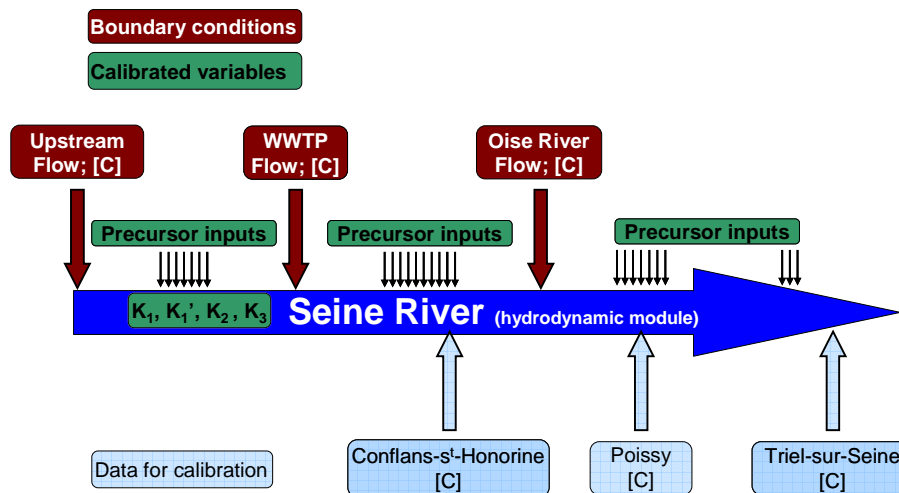
$$\frac{d[NP_1EC]}{dt} = +K_1' [NP_1EO] - K_2 [NP_1EC] + \text{precursor inputs} \quad \text{Equation 2}$$

$$\frac{d[4-NP]}{dt} = +K_1 [NP_1EO] + K_2 [NP_1EC] - K_3 [4-NP] \quad \text{Equation 3}$$

12  
13 Due to the lack of suitable information for  $NP_1EO$  degradation and in order to reduce the number of parameters  
14 to calibrate, we assume that the oxidation and the biodegradation of  $NP_1EO$  are equal ( $K_1 = K_1'$ ). Moreover, data  
15 on long chain nonylphenol ethoxylates are scarce and their attenuation rate constants are not determined. The  
16 semi-quantitative analyses reveal that the effluent of Seine Aval WWTP is from 2 ( $NP_3EO$ ) to 11 ( $NP_8EO$ ) times  
17 more concentrated than the upstream of the Seine River. In addition, by comparing the concentration equivalents  
18 found at Poissy to the expected ones (combining of Conflans-s<sup>t</sup>-Honorine and the Oise River), discrepancies of  
19 -39 % ( $NP_3EO$ ) to -64 % ( $NP_8EO$ ) are noticed. These discrepancies give evidence of the disappearance of these  
20 compounds in the Seine River due to biodegradation processes. In order to take into account the biodegradation  
21 of long chain ethoxylates within the Seine River, "precursor inputs" terms are considered in the ProSe model  
22 scheme as linear sources, also being calibrated during the calibration process (Equations 1 and 2). Based on the  
23 disappearance of long chain nonylphenol ethoxylate along the Seine River (revealed by semi-quantitative  
24 analyses), the precursor inputs are assumed to decrease along the simulated transect. The highest precursor  
25 inputs are estimated close to the Seine Aval discharge and decrease until Triel-sur-Seine following natural  
26 logarithm shape representative of first-order kinetics. These precursor inputs take into account the appearing of



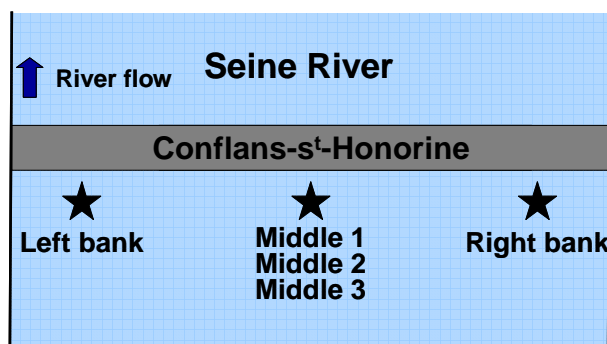
1 NP<sub>1</sub>EC or NP<sub>1</sub>EO along the studied transect due to the biodegradation of long chain compounds and they are  
 2 expressed in nanogram per liter of water and per day (ng/L/d).  
 3 The scheme in Fig. 3 shows the layout and input parameters of the model. In the ProSe model, it is necessary to  
 4 define the boundary conditions such as the upstream river flow (national discharge gauging station) and the  
 5 upstream compound concentrations [C] (field measurements). The attenuation rate constants and precursor  
 6 inputs are calibrated according to the concentrations at Conflans-S<sup>l</sup>-Honorine, Poissy and Triel-sur-Seine.



7  
 8 **Fig. 3** Scheme of the ProSe model and the required input parameters. ([C] = concentrations of 4 NP, NP<sub>1</sub>EO and  
 9 NP<sub>1</sub>EC in ng/L)  
 10

## 11 2.4. Sampling strategy for determining small scale spatial and temporal 12 variabilities

13 Small scale variabilities are crucial since they have a direct impact on the calibration process and the assessment  
 14 of the attenuation rate constants (Beven, 2010). The spatial and temporal variabilities of 4-NP, NP<sub>1</sub>EC and  
 15 NP<sub>1</sub>EO concentrations are assessed at Conflans-s<sup>l</sup>-Honorine (during September campaign) as depicted in Fig. 4  
 16 and are assumed to be representative of variabilities along the whole Seine River transect.



**Fig. 4** Sampling strategy to determine the small scale variabilities of concentrations. The sampling points are indicated by stars. Middle 1, 2 and 3 were sampled with an interval of 10 minutes.

For the assessment of the spatial variability, samplings are performed simultaneously at the left bank, in the middle and the right bank of the Seine River, while two additional samples in the middle are collected with a 10-min interval to assess the temporal variability. The relative standard deviations (%RSD) are used to assess the spatial (Left bank, Middle and Right bank) and temporal (Middle 1, 2 and 3) variabilities. The analytical repeatability is assessed by means of multiple extractions and UPLC-MS-MS analyses and the results are provided by Cladière *et al.*, (2013b) (4-NP: 7 % , NP<sub>1</sub>EC: 4 % and NP<sub>1</sub>EO: 14 %).

### 3. Results

#### 3.1. Small scale variabilities

Table 2 exhibits the concentrations of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO within the Seine River as well as small scale spatial and temporal variabilities.

**Table 2** Concentrations (ng/L) of left bank, middle and right bank of the Seine River and small scale variabilities

	Dissolved concentrations (ng/L)						Small scale variabilities (%)**		
	Spatial variability			Temporal variability			Spatial	Temporal	Total
	Left bank	Middle (average*)	Right bank	Middle 1	Middle 2	Middle 3			
4-NP	55	57	63	58	57	57	7	7	14
NP <sub>1</sub> EC	129	123	118	125	128	115	5	6	11
NP <sub>1</sub> EO	10	9	11	11	7	10	14	23	37

\*The concentrations for Middle used for the spatial variability are the averages of Middle 1, 2 and 3. \*\*The spatial and temporal variabilities are determined by the relative standard deviation of samples (%RSD). The total variabilities are calculated by the sum of spatial and temporal variabilites (we assume that analytical uncertainties are included in spatial and temporal variabilities).

The concentrations of 4-NP and NP<sub>1</sub>EC found in the Seine River ( $\approx 100$  ng/L) are far higher than their limits of quantification (4-NP: 24.5 ng/L; NP<sub>1</sub>EC: 1.7 ng/L) except for NP<sub>1</sub>EO (NP<sub>1</sub>EO: 9.8 ng/L). The repeatability is taken into account in the assessment of spatial and temporal variabilities by selecting the highest value between the relative standard deviation of samples and the analytical repeatability. The small scale variabilities (sum of spatial and temporal) of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO concentrations in the Seine River reach respectively 14 %,

1 11 % and 37 %. These results provide quantification errors due to the sampling protocol, which are usually not  
 2 reported in most of articles dealing with pollutants in receiving surface water.

3 The sampling strategy is thus applicable to 4-NP and NP<sub>1</sub>EC measurements, but is less accurate for NP<sub>1</sub>EO.  
 4 Therefore, the variabilities on Seine River profiles are plotted as error bars for the measured concentrations. In  
 5 addition, the variabilities on boundary conditions are considered on the modelled profiles.

### 6 3.2. Attenuation rate constants

7 The concentrations of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO found in the Seine and Oise Rivers and the effluents of Seine  
 8 Aval WWTP as well as the biogeochemical parameters (water temperature, dissolved organic carbon (DOC),  
 9 dissolved oxygen and pH) are exhibited in Table 3. All concentrations found during July and September  
 10 sampling campaigns are consistent with the literature and previous results found in the Seine River (Jonkers *et*  
 11 *al.*, 2009; Loos *et al.*, 2010; Cladière *et al.*, 2013b). The concentrations found during July in the Seine River are  
 12 higher than those found during September, mainly due to the significant higher concentrations of Seine Aval  
 13 WWTP and the Oise River (Table 3). Whatever the sampling campaign, no exceedance of the European Quality  
 14 Standard (300 ng/L) is observed along the 28 km transect of the Seine River even downstream of Seine Aval  
 15 WWTP.

16 **Table 3** Biogeochemical parameters and nonylphenolic compound concentrations measured in July and  
 17 September sampling campaigns

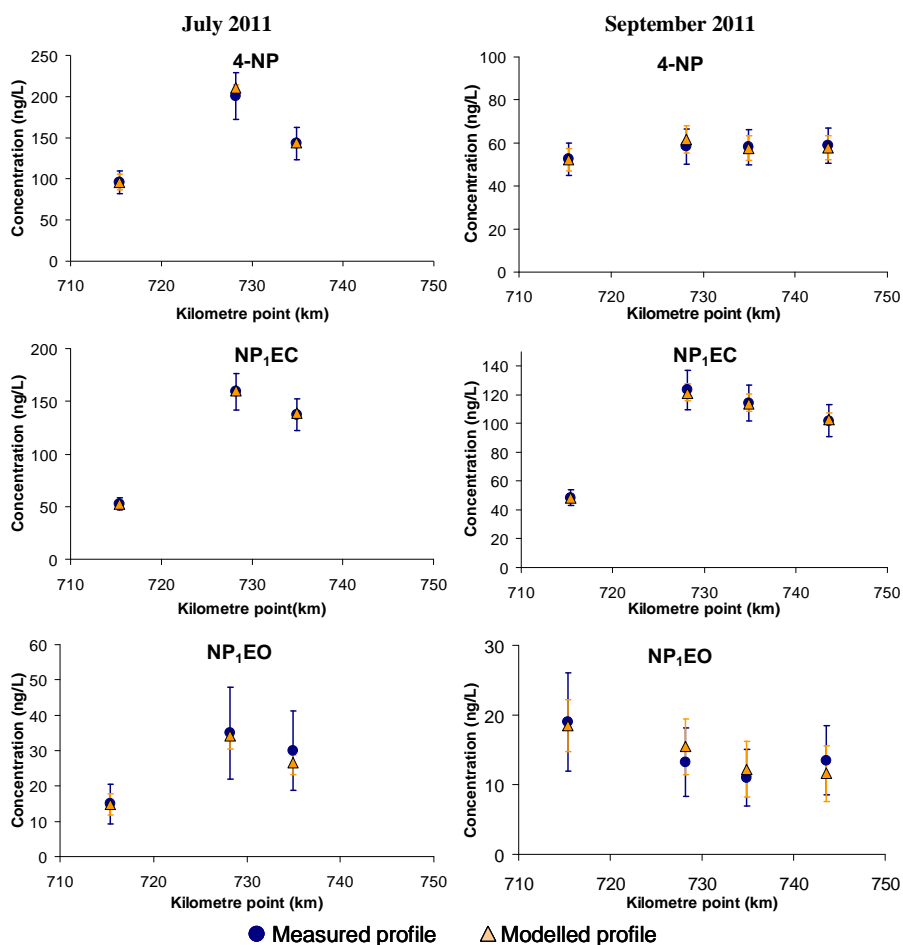
Parameters*	July					September					
	Upstream	Seine Aval WWTP	Conflans-st-Honorine	Oise river	Poissy	Upstream	Seine Aval WWTP	Conflans-st-Honorine	Oise river	Poissy	Triel-sur-Seine
Flow (m <sup>3</sup> /s)	72	17	89	32	121	97	18	115	28	143	143
Tw (°C)	20.8	20.9	22.2	21.4	21.0	19.3	ND	19.1	17.5	20.8	ND
σ (μS/cm)	600	1195	693	690	699	612	720	635	739	666	ND
pH	8.0	ND**	7.5	7.8	7.6	7.6	ND	7.6	8.0	7.6	ND
O <sub>2</sub> (mg/L)	4.9	ND	4.0	5.0	ND	ND	ND	9.3	11.3	7.7	ND
DOC (mg/L)	2.95	11.60	4.08	2.74	3.83	2.84	8.97	3.74	3.32	3.37	3.63
4-NP (ng/L)	102	320	201	138	143	53	136	57	35	45	70
NP <sub>1</sub> EC (ng/L)	52	751	159	66	137	49	616	123	76	114	102
NP <sub>1</sub> EO (ng/L)	19	80	35	17	30	19	43	9	20	11	14

18 \*σ: conductivity, O<sub>2</sub>: dissolved oxygen, DOC: dissolved organic carbon. \*\*ND: not determined.

19  
 20 Before calibrating the attenuation rate constants in the Seine River, a first simulation is performed considering  
 21 nonylphenolic compounds as a conservative tracer for July and September campaigns. Results (data not shown)  
 22 underscore that the modelled trends do not fit the observed ones indicating that only physical processes  
 23 (transport and lateral inflows) do not explain the fate of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO along the Seine River.

1 Attenuation rate constants and precursor inputs are, thereafter, calibrated (Fig 5). In parallel, small scale  
 2 variability intervals are used to assess uncertainties around the optimal values of attenuation rate constants.

3



4 **Fig. 5** Modelled (optimised rate constants) and actual profiles of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO concentrations in the  
 5 Seine River. The points exhibited are Maisons-Laffite (715.4 km), Conflans-st-Honorine (728.2 km),  
 6 Poissy (734.9 km) and Triel-sur-Seine (743.6 km). The y error bars represented on the actual profile  
 7 indicate the total variabilities linked to in-situ sampling while they represent the uncertainties attached to  
 8 boundary conditions on modelled profile  
 9  
 10

11 Table 4 shows the minimum, optimal and maximum attenuation rate constants as well as precursor inputs for  
 12 July and September sampling campaigns.

13

1 **Table 4** Attenuation rate constants ( $d^{-1}$ ) and precursor inputs (ng/L/d)

	Attenuation rate constants ( $d^{-1}$ )			Precursor inputs (ng/L/d)	
	$K_1 = K_1'$	$K_2$	$K_3$	NP <sub>1</sub> EC	NP <sub>1</sub> EO
	min - opt - max	min - opt - max	min - opt - max	min - max	min - max
July 2011	0.05 - 0.10 - 0.15	3.14 - 3.30 - 3.47	2.38 - 2.50 - 2.75	3.44 - 7.84	0.06 - 0.4
September 2011	0.29 - 0.30 - 0.33	0.08 - 0.10 - 0.14	0.09 - 0.15 - 0.19	0.01 - 0.3	0.04 - 0.4
Jonkers <i>et al.</i> , 2005 (estuary)	0.060 - 0.089	0.019 - 0.020	0.024 - 0.043	-	-
Jonkers <i>et al.</i> , 2005 (estimation river)	0.18 - 0.21	0.048 - 0.159	0.072 - 0.21	-	-
Staples <i>et al.</i> , 2001 (laboratory)	0.065	0.099	0.076 - 0.092	-	-

2 Min and max values were assessed according to the small scale variabilities. Optimised values (opt) enable the best fit between simulated  
3 and observed profiles.

4  
5 For July, Table 4 discloses high values for  $K_2$  (3.14 to 3.47  $d^{-1}$ ) and  $K_3$  (2.38 to 2.75  $d^{-1}$ ). These rate constants are  
6 far higher than those reported by Jonkers *et al.*, (2005) in the Rhine estuary and Staples *et al.*, (2001) in  
7 laboratory batch reactor (Table 4). In parallel, the precursor inputs were estimated between 7.84 ng/L/d and  
8 3.44 ng/L/d for NP<sub>1</sub>EC and between 0.4 ng/L/d and 0.06 ng/L/d for NP<sub>1</sub>EO. In the case of NP<sub>1</sub>EC, the total  
9 precursor flux along the Seine River transect (33.6 g/d) accounts for 10 % of the upstream load (370 g/d) and  
10 only for 2.4 % of the Seine Aval input (1,480 g/d) which means that they are not main sources of NP<sub>1</sub>EC and  
11 NP<sub>1</sub>EO into the Seine River. However, the precursors inputs are crucial in the fate of NP<sub>1</sub>EC and NP<sub>1</sub>EO since  
12 neglect them during the calibration process does not allow a good fit between modelled and observed profiles.  
13 Calibration parameters for the September campaign are much lower (Table 4). Mean values of  $K_2$  and  $K_3$  are  
14 respectively 17 and 33 times lower than those estimated for the July campaign. In parallel, the maximum  
15 precursor inputs for NP<sub>1</sub>EC drops by a factor 26. The decreases of  $K_2$ ,  $K_3$  and NP<sub>1</sub>EC precursor inputs suggest  
16 that the aerobic biodegradation was significantly higher in July compared to September. Moreover, the rate  
17 constants estimated in September are consistent with the literature (Jonkers *et al.*, 2005; Staples *et al.*, 2001).  
18 The July and September sampling campaigns enable to calibrate of attenuation rate constants but not to validate  
19 them. Validation of the parameter sets is done at the annual scale based on 11 monthly sampling campaigns  
20 performed at the Maisons-Laffitte, Seine Aval WWTP effluent, the Oise River and Meulan (downstream Triel-  
21 sur-Seine). The validation is provided in the companion paper: "Modelling the fate of nonylphenolic compounds  
22 in the Seine River - part 2: assessing the impact of global change on daily concentrations" (Cladière *et al.*  
23 2013a).

## 4. Discussion

### 4.1. Influence of biogeochemical conditions of the Seine River

In order to understand the significant differences between July and September, the biogeochemical conditions of the Seine River are examined (e.g. temperature, pH, chlorophyll a, dissolved O<sub>2</sub>, Seine River flow). Manzano *et al.*, (1999) highlight the impact of temperature on biodegradation rates of nonylphenolic compounds and report that the higher the temperature, the higher the biodegradation rate. Water temperature is similar in July (21°C) and September (20°C) and cannot explain such differences of the attenuation rate constants. Furthermore, pH, conductivity, meteorological and hydrological conditions of the Seine River were quite similar during both campaigns.

Chlorophyll a concentrations are continuously monitored in the middle of the considered transect and exhibit high values (up to 20 µg/L representative of an algal bloom in the Seine River) one week before the July sampling campaign (data not shown). On the contrary, during September, low concentrations of chlorophyll a ( $\approx 5$  µg/L) are noticed.

The link between an algal bloom and the growth of heterotrophic bacteria has been reported by Kisand and Noges, (1998) in lake Vortsjarv (Estonia) and by Hygum *et al.*, (1997) for laboratory experiments. Both studies reported a significant increase of the bacteria growth during the decline of an algal bloom. The high activity of heterotrophic bacteria is also visible on dissolved O<sub>2</sub> since the concentrations measured in July are close to 4 mg/L and saturation about 75 % while in September the concentration are far higher at 9 mg/L and saturation at 100 % (Table 3). Based on this conclusion, the decline of the algal bloom during the July campaign favours the increase of heterotrophic bacterial biomass, and consequently enhances the biodegradation of organic matters such as nonylphenolic compounds since Corvini *et al.*, (2006) reported that various environmental bacteria species can biodegrade them. Thus, in July the half-life times of 4-NP and NP<sub>1</sub>EC vary between 5 and 7 hours, while in September they are longer and reach 7 days. The half-life time of NP<sub>1</sub>EO remains constant in July and September and is close to 2 days.

Finally, the differences between July and September campaigns suggest that the bacterial biomass of the Seine River influences the oxidative biodegradation of NPnEO (K<sub>2</sub>, K<sub>3</sub>, NP<sub>1</sub>EC precursors inputs), especially after an algal bloom. The first-order kinetic approach used in this study seems reliable to describe a punctual state of biodegradation in the Seine River (constant bacterial biomass) but does not take into account the variabilities generated by the fluctuation of bacterial biomass. In order to clarify the impacts of heterotrophic bacteria on

1 nonylphenolic compounds biodegradation, further studies should focus on coupling the modelling of NPnEO  
 2 biodegradation and bacterial biomass using biogeochemical models such as ProSe.  
 3 In contrast, the non oxidative biodegradation of NPnEO is not impacted by the heterotrophic bacterial biomass of  
 4 the Seine River since the attenuation rate constants of NP<sub>1</sub>EO do not significantly evolve and the NP<sub>1</sub>EO  
 5 precursor inputs remain constant between both campaigns. However, this result must be considered carefully  
 6 according to the lower confidence on NP<sub>1</sub>EO concentration profiles (small scale variability = 37 % and  
 7 concentrations close to the quantification limit).

## 8 **4.2. Sensitivity analysis**

9 In order to assess the significance of each process in the biodegradation pathway, a sensitivity analysis is  
 10 performed using an approach “one factor at a time” (OFAT) (Félix and Xanthoulis, 2005). Input parameters of  
 11 the model are modified by -10 % and +10 % around its optimised value one after the other. The effect of each  
 12 modification is analyzed on the outputs by the means of a sensitivity index (SI, Equation 4).

$$SI = \frac{\frac{O_{Test} - O_{Opt}}{O_{mean}}}{\frac{I_{Test} - I_{Opt}}{I_{mean}}} \quad \text{Equation 4}$$

13 Where

14 SI is the sensitivity index;

15 I<sub>Opt</sub> is the optimised value of input;

16 I<sub>Test</sub> is the tested value of input (+10 % or -10 %);

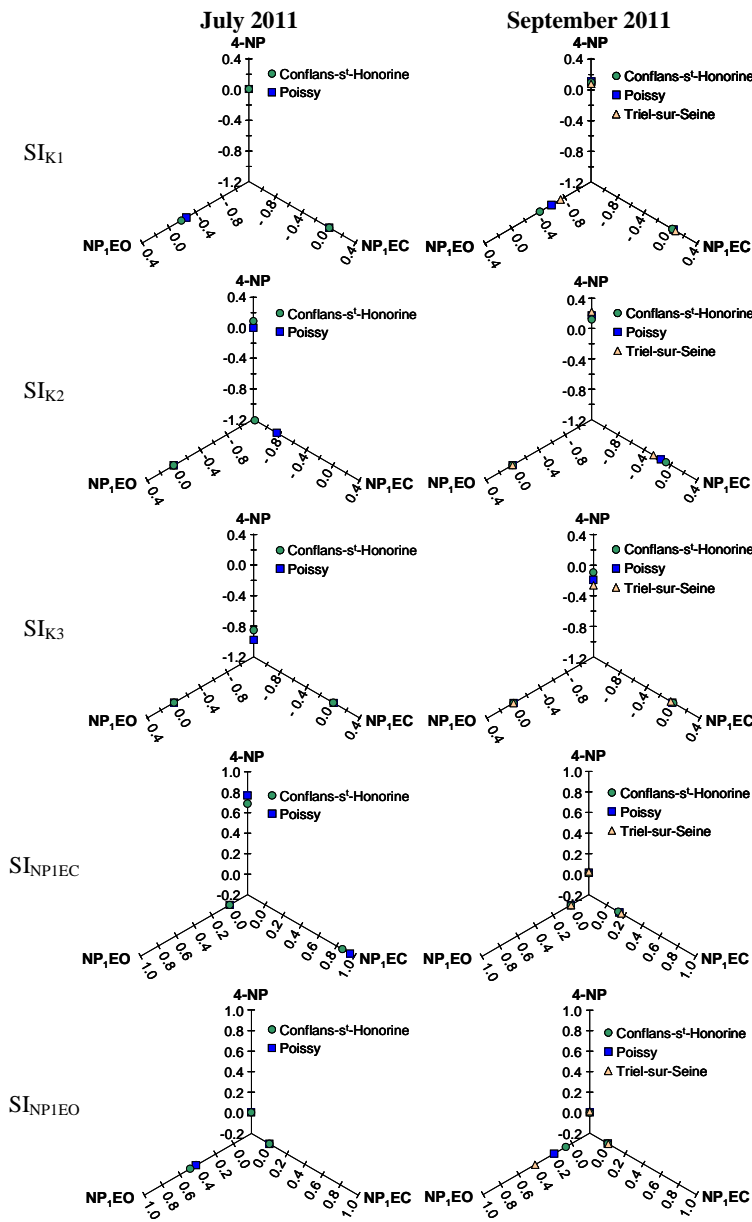
17 I<sub>mean</sub> is the mean of I<sub>opt</sub> and I<sub>Test</sub>;

18 O<sub>Test</sub> and O<sub>Opt</sub> are the outputs respectively related to I<sub>Test</sub> and I<sub>Opt</sub>;

19 O<sub>mean</sub> is the mean of O<sub>Test</sub> and O<sub>opt</sub>.

20 The sensitivity index weights the impacts of input parameters such as attenuation rate constants or precursor  
 21 inputs on model outputs (concentrations of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO at Conflans-s<sup>t</sup>-Honorine, Poissy and Triel-  
 22 sur-Seine). A positive SI means that input and outputs vary in a similar way while a negative SI means that  
 23 inputs and outputs vary inversely. In addition, the higher the absolute value of SI, the higher the impact of input  
 24 parameter on the considered output. The sensitivity indexes of nonylphenolic compound concentrations in the  
 25 Seine River towards attenuation rate constants (SI<sub>K1</sub>, SI<sub>K1'</sub>, SI<sub>K2</sub> and SI<sub>K3</sub>) and precursor inputs (SI<sub>NP<sub>1</sub>EO</sub>, SI<sub>NP<sub>1</sub>EC</sub>)

1 are shown in Fig. 6. During this study, an increase or a decrease of the considered inputs give similar results of  
 2 sensitivity and hence results are compiled in Fig.6.



3  
 4 **Fig. 6** Sensitivity indexes of simulated concentrations of 4-NP, NP<sub>1</sub>EO and NP<sub>1</sub>EC at Conflans-st-Honorine,  
 5 Poissy and Triel-sur-Seine towards attenuation rate constants and precursor inputs  
 6

7 Whatever the sampling campaign,  $SI_{K1(K1')}$ ,  $SI_{K2}$  and  $SI_{K3}$  are negative meaning that an increase of attenuation  
 8 rate constants leads to a decrease of modelled concentrations. As underscored in the Fig.6, the sensitivity indexes  
 9 of biodegradation processes in July and September are not similar confirming that campaigns occur in different  
 10 conditions.

11 In July, high values of  $SI_{K2}$  (varying from -0.8 - to -1.2) and  $SI_{K3}$  (ranging from -0.8 to -1.0) are noticed for  
 12 NP<sub>1</sub>EC and 4-NP concentrations. On the contrary, the  $SI_{K1}$  (from -0.18 to -0.26) for NP<sub>1</sub>EO concentrations is



1 smaller but still significant. In September, the  $SI_{K_2}$  and  $SI_{K_3}$  are smaller (around -0.25 for NP<sub>1</sub>EC and 4-NP  
2 concentrations). This observation confirms that the aerobic biodegradation process is important in July while its  
3 significance is lower in September. These results corroborate the hypothesis of an enhanced aerobic  
4 biodegradation at the end of an algal bloom. A similar conclusion is highlighted for  $SI_{NP_{1EC}}$  between July  
5 ( $SI_{NP_{1EC}} \approx 1$ ) and September ( $SI_{NP_{1EC}} \approx 0.2$ ).

6 Despite a significant role of  $K_2$  to determine NP<sub>1</sub>EC concentrations, it has a very limited impact on 4-NP  
7 concentrations (biodegradation product), as confirmed by the  $SI_{K_2}$  for 4-NP concentrations (0.09 in July and  
8 September). A similar result is noticed for  $SI_{K_1(K_1')}$  for NP<sub>1</sub>EC or 4-NP concentrations ( $\approx 0.007$ ). As a  
9 conclusion, according to these sensitivity indexes, a variation of attenuation rate constants  $K_1$  and  $K_2$  has a weak  
10 influence on 4-NP concentrations along the simulated transect and points out the need to extend the scale of  
11 modelling (spatial and temporal) to better understand the impact of biodegradation processes.

## 12 **5. Conclusions**

13 This study assesses the *in-situ* attenuation rate constants of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO within surface water in a  
14 heavily urbanized area such as Paris. Based on a coupling between sampling campaigns and hydraulic and  
15 biogeochemical simulations, this study provides the first real case study of the environmental fate of 4-NP,  
16 NP<sub>1</sub>EC and NP<sub>1</sub>EO in freshwater. This study is of prime interest since it proves the possibility to simulate the  
17 fate of readily biodegradable pollutants such as 4-NP and its precursors within surface water. Based on the  
18 toxicity of these compounds, the assessment of their environmental fate is a key issue to understand their  
19 persistence and their possible impacts on aquatic wildlife. Results highlight a strong temporal variability of rate  
20 constants between July and September campaigns. While the rate constants evaluated for July are higher than  
21  $1 \text{ d}^{-1}$  and NP<sub>1</sub>EC precursor inputs higher than 344 ng/L/d, the rate constants and precursor inputs assessed for  
22 September are smaller ( $K \approx 0.1 \text{ d}^{-1}$  and precursors inputs  $\approx 2 \text{ ng/L/d}$ ) but consistent with the literature. The  
23 variability of rate constants seems to be linked to the heterotrophic bacterial biomass and highlights that the first-  
24 order kinetic approach used in this study is reliable to describe punctual conditions of the Seine River (with a  
25 constant bacterial biomass) but not to describe the complexity of biogeochemical processes at larger temporal  
26 scale. Consequently, the coupling between attenuation rate constants and bacterial biomass should be more  
27 intensively investigated to definitely validate the impact of biogeochemical conditions of surface water on  
28 biodegradation of nonylphenolic compounds.

1 The sensitivity analysis of the ProSe model points out that the aerobic biodegradation processes are important  
2 inputs for the ProSe model and reinforces the hypothesis of the large influence of the biogeochemical conditions  
3 of the Seine River. However, this sensitivity analysis also highlights that biodegradation processes are not a key  
4 input for 4-NP fate along the 28 km transect, and points out the need to lengthen the transect in order to confirm  
5 or contradict this last conclusion at the basin scale.

6 In order to validate the modelling parameter values, the attenuation rate constants assessed during both sampling  
7 campaigns will be applied to simulate daily of concentrations of 4-NP, NP<sub>1</sub>EC and NP<sub>1</sub>EO for 2010 and  
8 compared to monthly sampling campaigns. To this end, 11 sampling campaigns were carried out from February  
9 to December 2010 at Maisons-Laffite, Seine Aval WWTP effluent, the Oise River, and Meulan (downstream of  
10 Triel-sur-Seine).

11 At last, the impact of global changes (global warming, population growth, optimisation of WWTP) on daily  
12 concentrations will be studied for the middle and the late 21<sup>st</sup> century. This is presented in a companion paper  
13 entitled: “Modelling the fate of nonylphenolic compounds in the Seine River - part 2: assessing the impact of  
14 global change on daily concentrations” (Cladière *et al.*, 2013a).

## 15 **6. Acknowledgments**

16 The authors want to gratefully acknowledge the Paris public sanitation service (SIAAP) and especially Vincent  
17 Rocher and Céline Briand for their valuable assistance for sampling the effluent of Seine Aval WWTP. The  
18 authors kindly thank Alexis Groleau and Nicolas Escoffier, who provided access to the 15min-chloropyll a  
19 monitoring data of the Seine River. The authors also acknowledge Mohamed Saad for his technical assistance.

20 This study was supported by the PIREN-Seine research programme.

21 We thank Science of the Total Environment to permit us to publish the post-print of the manuscript.

## 22 **7. References**

- 23 Ahel M, Giger W, Schaffner C. Behaviour of alkylphenol polyethoxylate surfactants in the aquatic environment-  
24 -II. Occurrence and transformation in rivers. *Water Research* 1994; 28: 1143-1152.
- 25 Ayorinde FO, Eribo BE, Johnson JH, Elhilo E. Molecular distribution of some commercial nonylphenol  
26 ethoxylates using matrix-assisted laser desorption ionization time-of-flight mass spectrometry. *Rapid*  
27 *Communications in Mass Spectrometry* 1999; 13: 1124-1128.
- 28 Beven KJ. Preferential flows and travel time distributions: Defining adequate hypothesis tests for hydrological  
29 process models. *Hydrological Processes* 2010; 24: 1537-1547.
- 30 Cladière M, Bonhomme C, Vilmin L, Gasperi J, Flipo N, Habets F, Tassin B. Modelling the fate of  
31 nonylphenolic compounds in the Seine River - part 2: assessing the impact of global changes on annual  
32 time series. *Science of the Total Environment* 2013a; (submitted STOTEN-D-13-02231).

- 1 Cladière M, Gasperi J, Gilbert S, Lorgeoux C, Tassin B. Alkylphenol ethoxylates and bisphenol A in surface  
2 water within a heavily urbanized area, such as Paris. In: Marinov AM, Brebbia CA, editors. Water  
3 Pollution X. 135. Wit Press, Southampton, 2010, pp. 131-142.
- 4 Cladière M, Gasperi J, Lorgeoux C, Bonhomme C, Rocher V, Tassin B. Alkylphenolic compounds and  
5 bisphenol A contamination within a heavily urbanized area: case study of Paris. *Environmental Science  
6 and Pollution Research* 2013b; 20: 2973 - 2983.
- 7 Corvini PFX, Schaffer A, Schlosser D. Microbial degradation of nonylphenol and other alkylphenols - our  
8 evolving view. *Applied Microbiology and Biotechnology* 2006; 72: 223-243.
- 9 European Commission. Directive 2000/60/EC of the European Parliament and of the Council of 23 October  
10 2000 establishing a framework for Community action in the field of water policy, Official journal of the  
11 European Union n° 327, 2000, pp. 1-72.
- 12 European Commission. Directive 2008/105/EC of the European Parliament and of the Council of 16 December  
13 2008 on environmental quality standards in the field of water policy, amending and subsequently  
14 repealing Council Directives 82/176/EEC, 83/513/EEC, 84/156/EEC, 84/491/EEC, 86/280/EEC and  
15 amending Directive 2000/60/EC of the European Parliament and of the Council. European  
16 Commission, Official Journal of the European Union, 2008, pp. 84-97.
- 17 Even S, Billen G, Bacq N, Théry S, Ruelland D, Garnier J, et al. New tools for modelling water quality of  
18 hydrosystems: An application in the Seine River basin in the frame of the Water Framework Directive.  
19 *Science of the Total Environment* 2007; 375: 274 - 291.
- 20 Even S, Poulin M, Gamier J, Billen G, Servais P, Chesterikoff A, et al. River ecosystem modelling: Application  
21 of the PROSE model to the Seine river (France). *Hydrobiologia* 1998; 373-374: 27-45.
- 22 Félix R, Xanthoulis D. Sensitivity analysis of the mathematical model "Erosion Productivity Impact Calculator"  
23 (EPIC) by approach One-Factor-At-A-Time (OAT). *Biotechnology, Agronomy, Society and  
24 Environment International* 2005; 9: 179-190.
- 25 Fenet H, Gomez E, Pillon A, Rosain D, Nicolas JC, Casellas C, et al. Estrogenic activity in water and sediments  
26 of a French river: Contribution of alkylphenols. *Archives of Environmental Contamination and  
27 Toxicology* 2003; 44: 1-6.
- 28 Flipo, N., Rabouille, C., Poulin, M., Even, S., Tusseau-Vuillemin, M., Lalande, M. Primary production in  
29 headwater streams of the Seine basin: the Grand Morin case study. *Science of the Total Environment*  
30 2007; 375: 98-109.
- 31 Gabriel FLP, Heidlberger A, Rentsch D, Giger W, Guenther K, Kohler HPE. A novel metabolic pathway for  
32 degradation of 4-nonylphenol environmental contaminants by *Sphingomonas xenophaga* Bayram -  
33 ipso-hydroxylation and intramolecular rearrangement. *Journal of Biological Chemistry* 2005; 280:  
34 15526-15533.
- 35 Giger W, Brunner PH, Schaffner C. 4-nonylphenol in sewage sludge - Accumulation of toxic metabolites from  
36 nonionic surfactants. *Science* 1984; 225: 623-625.
- 37 Giger W, Gabriel FLP, Jonkers N, Wettstein FE, Kohler H-PE. Environmental fate of phenolic endocrine  
38 disruptors: field and laboratory studies. *Philosophical Transactions of the Royal Society A:  
39 Mathematical, Physical and Engineering Sciences* 2009; 367: 3941-3963.
- 40 Hygum BH, Petersen JW, Sondergaard M. Dissolved organic carbon released by zooplankton grazing activity -  
41 A high-quality substrate pool for bacteria. *Journal of Plankton Research* 1997; 19: 97-111.
- 42 John DM, White GF. Mechanism for Biotransformation of Nonylphenol polyethoxylates to xenoestrogens in  
43 *Pseudomonas putida*. *Journal of Bacteriology* 1998; 180: 4332-4338.
- 44 Jonkers N, Knepper TP, De Voogt P. Aerobic biodegradation studies of nonylphenol ethoxylates in river water  
45 using liquid chromatography-electrospray tandem mass spectrometry. *Environmental Science &  
46 Technology* 2001; 35: 335-340.
- 47 Jonkers N, Kohler HPE, Dammshäuser A, Giger W. Mass flows of endocrine disruptors in the Glatt River during  
48 varying weather conditions. *Environmental Pollution* 2009; 157: 714-723.
- 49 Jonkers N, Laane RWPM, de Graaf C, de Voogt P. Fate modeling of nonylphenol ethoxylates and their  
50 metabolites in the Dutch Scheldt and Rhine estuaries: validation with new field data. *Estuarine, Coastal  
51 and Shelf Science* 2005; 62: 141-160.
- 52 Jugan ML, Oziol L, Bimbot M, Huteau V, Tamisier-Karolak S, Blondeau JP, et al. In vitro assessment of thyroid  
53 and estrogenic endocrine disruptors in wastewater treatment plants, rivers and drinking water supplies  
54 in the greater Paris area (France). *Science of The Total Environment* 2009; 407: 3579-3587.
- 55 Jurado E, Fernandez-Serrano M, Nunez-Olea J, Lechuga M. Aerobic Biodegradation of a Nonylphenol  
56 Polyethoxylate and Toxicity of the Biodegradation Metabolites. *Bulletin of Environmental  
57 Contamination and Toxicology* 2009; 83: 307-312.
- 58 Karahan O, Olmez-Hanci T, Arslan-Alaton I, Orhon D. Modelling biodegradation of nonylphenol ethoxylate in  
59 acclimated and non-acclimated microbial cultures. *Bioresource Technology* 2010; 101: 8058-8066.

- 1 Kisand V, Noges T. Seasonal dynamics of bacterio- and phytoplankton in large and shallow eutrophic Lake  
2 Vortsjarv, Estonia. *International Review of Hydrobiology* 1998; 83: 205-216.
- 3 Loos R, Locoro G, Contini S. Occurrence of polar organic contaminants in the dissolved water phase of the  
4 Danube River and its major tributaries using SPE-LC-MS2 analysis. *Water Research* 2010; 44: 2325-  
5 2335.
- 6 Manzano MA, Perales JA, Sales D, Quiroga JM. The effect of temperature on the biodegradation of a  
7 nonylphenol polyethoxylate in river water. *Water Research* 1999; 33: 2593-2600.
- 8 Montgomery-Brown J, Li YM, Ding WH, Mong GM, Campbell JA, Reinhard M. NP1EC degradation pathways  
9 under oxic and microxic conditions. *Environmental Science & Technology* 2008; 42: 6409-6414.
- 10 Petrovic M, Eljarrat E, Lopez De Alda MJ, Barcelo D. Endocrine disrupting compounds and other emerging  
11 contaminants in the environment: A survey on new monitoring strategies and occurrence data.  
12 *Analytical and Bioanalytical Chemistry* 2004; 378: 549-562.
- 13 Soto AM, Justicia H, Wray JW, Sonnenschein C. Para-nonyl-phenol - An estrogenic xenobiotic released from  
14 modified polystyrene. *Environmental Health Perspectives* 1991; 92: 167-173.
- 15 Staples CA, Naylor CG, Williams JB, Gledhill WE. Ultimate biodegradation of alkylphenol ethoxylate  
16 surfactants and their biodegradation intermediates. *Environmental Toxicology and Chemistry* 2001; 20:  
17 2450-2455.
- 18 Ying G-G, Williams B, Kookana R. Environmental fate of alkylphenols and alkylphenol ethoxylates--a review.  
19 *Environment International* 2002; 28: 215-226.